

# Comparisons of watershed sulfur budgets in southeast Canada and northeast US: new approaches and implications

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Received: 21 December 2009 / Accepted: 18 April 2010  
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**Abstract** Most of eastern North America receives elevated levels of atmospheric deposition of sulfur (S) that result from anthropogenic SO<sub>2</sub> emissions from fossil fuel combustion. Atmospheric S deposition has acidified sensitive terrestrial and aquatic ecosystems in this region; however, deposition has been declining since the 1970s, resulting in some

recovery in previously acidified aquatic ecosystems. Accurate watershed S mass balances help to evaluate the extent to which atmospheric S deposition is retained within ecosystems, and whether internal cycling sources and biogeochemical processes may be affecting the rate of recovery from decreasing S atmospheric loads. This study evaluated S mass balances for 15 sites with watersheds in southeastern Canada and northeastern US for the period 1985 to 2002. These 15 sites included nine in Canada (Turkey Lakes, ON; Harp Lake, ON; Plastic Lake, ON; Hermine, QC; Lake Laflamme, QC; Lake Clair, QC; Lake Tirasse, QC; Mersey, NS; Moosepit, NS) and

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**Electronic supplementary material** The online version of this article (doi:10.1007/s10533-010-9455-0) contains supplementary material, which is available to authorized users.

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six in the US (Arbutus Lake, NY; Biscuit Brook, NY; Sleepers River, VT; Hubbard Brook Experimental Forest, NH; Cone Pond, NH; Bear Brook Watershed, ME). Annual S wet deposition inputs were derived from measured bulk or wet-only deposition and stream export was obtained by combining drainage water fluxes with  $\text{SO}_4^{2-}$  concentrations. Dry deposition has the greatest uncertainty of any of the mass flux calculations necessary to develop accurate watershed balances, and here we developed a new method to calculate this quantity. We utilized historical information from both the US National Emissions Inventory and the US (CASTNET) and the Canadian (CAPMoN) dry deposition networks to develop a formulation that predicted  $\text{SO}_2$  concentrations as a function of  $\text{SO}_2$  emissions, latitude and longitude. The  $\text{SO}_2$  concentrations were used to predict dry deposition using relationships between concentrations and deposition flux derived from the CASTNET or CAPMoN networks. For the year 2002, we compared the  $\text{SO}_2$  concentrations and deposition predictions with the predictions of two continental-scale air quality models, the Community Multiscale Air Quality (CMAQ) model and A Unified Regional Air-quality Modeling System (AURAMS) that utilize complete inventories of emissions and chemical budgets. The results of this

comparison indicated that the predictive relationship provides an accurate representation of  $\text{SO}_2$  concentrations and S deposition for the region that is generally consistent with these models, and thus provides confidence that our approach could be used to develop accurate watershed S budgets for these 15 sites. Most watersheds showed large net losses of  $\text{SO}_4^{2-}$  on an annual basis, and the watershed mass balances were grouped into five categories based on the relative value of mean annual net losses or net gains. The net annual fluxes of  $\text{SO}_4^{2-}$  showed a strong relationship with hydrology; the largest net annual negative fluxes were associated with years of greatest precipitation amount and highest discharge. The important role of catchment hydrology on S budgets suggests implications for future predicted climate change as it affects patterns of precipitation and drought. The sensitivity of S budgets is likely to be greatest in watersheds with the greatest wetland area, which are particularly sensitive to drying and wetting cycles. A small number of the watersheds in this analysis were shown to have substantial S sources from mineral weathering, but most showed evidence of an internal source of  $\text{SO}_4^{2-}$ , which is likely from the mineralization of organic S stored from decades of increased S deposition. Mobilization of this internal S appears to contribute about

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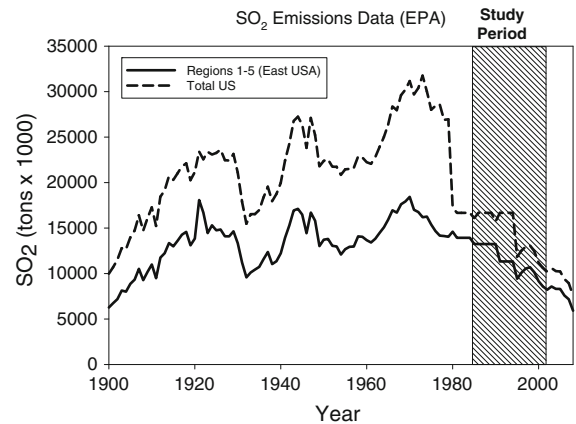
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1–6 kg S ha<sup>-1</sup> year<sup>-1</sup> to stream fluxes at these sites and is affecting the rate and extent of recovery from acidification as S deposition rates have declined in recent years. This internal S source should be considered when developing critical deposition loads that will promote ecosystem recovery from acidification and the depletion of nutrient cations in the northeastern US and southeastern Canada.

**Keywords** Watersheds · Sulfur budgets · Atmospheric deposition models · Acidic deposition · Recovery · Northeast US · Southeast Canada

## Introduction

Anthropogenic emissions of sulfur dioxide (SO<sub>2</sub>) in North America have shown marked temporal changes over the past 100 years with a maximum occurring in the early 1970s followed by a substantial decline (Fig. 1). Much of this decline in the US was driven by the enactment of the Clean Air Act (CAA) in 1970 and subsequent Title IV Amendment of the CAA in 1990 as well as other regulatory controls on SO<sub>2</sub> emissions. Similarly, implementation of the Eastern Canada Acid Rain Program reduced Canadian emissions such that total Canada-US SO<sub>2</sub> emissions were 14 million tonnes in 2006—a 50% reduction relative to 1980 levels (Canada-US 1992, 2008). These historical emission trends are matched by changes in the atmospheric concentration and deposition of S (Lynch et al. 1996; Sickles and Shadwick 2007; Weathers et al. 2006a). Spatially extensive and



**Fig. 1** SO<sub>2</sub> emissions for entire United States and Regions 1–5 (eastern U.S.) from 1900 to 2008. Source of information is provided in the text

quantitatively constrained results showing these changes have been documented for wet deposition, but decreases have also been noted for components of dry deposition (composed mostly of SO<sub>2</sub> and SO<sub>4</sub><sup>2-</sup> aerosols; Baumgardner et al. 2002). These decreases in atmospheric S inputs have also resulted in decreases in SO<sub>4</sub><sup>2-</sup> concentrations in surface waters with notable decreases across southeastern Canada (Clair et al. 1995; Jeffries et al. 2003a, b; Houle et al. 2004) and the northeastern US (Stoddard et al. 1999, 2003; Likens et al. 2002, 2005; Martin et al. 2000; Driscoll et al. 2003).

Elevated S deposition has been closely linked with the acidification of soils and surface waters (Reuss and Johnson 1986; Likens et al. 1996; Weathers and Lovett 1998). This acidification has resulted in the mobilization of toxic cations (e.g., aluminum) (Driscoll and Postek 1996) and the depletion of soil nutrient cations (e.g., Ca<sup>2+</sup>, Mg<sup>2+</sup>) (Likens et al. 1996, 1998; Fernandez et al. 2003; Bailey et al. 2005; Sullivan et al. 2006). Changes in SO<sub>4</sub><sup>2-</sup> concentrations have been linked to effects of microbial dissimilatory sulfate reduction, an increase in methane production (Gauci et al. 2008), and the methylation of mercury (Gilmour et al. 1992). Increased concentrations of atmospheric methane are important due to the high heat trapping capacity of this “greenhouse gas” (IPCC 2001). Methylmercury is bioaccumulated along food chains, and this chemical form of mercury is highly toxic to biota, including humans (e.g., Dennis et al. 2005).

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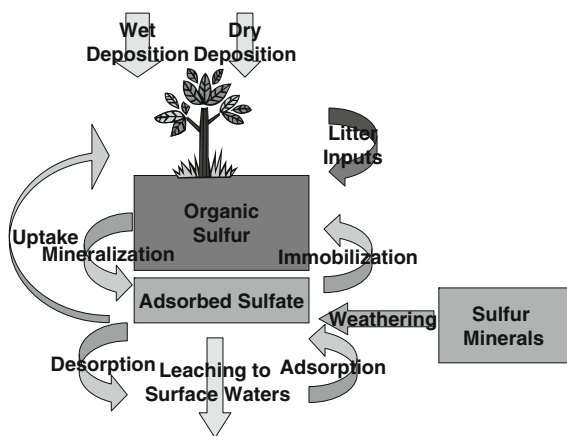
## Sulfur pools and budgets

Previous studies documented and summarized sulfur (S) budgets in forested ecosystems and their respective watersheds in Europe and North America. European studies have emphasized the dramatic changes in S budgets associated with those regions that had been subjected to large reductions in S deposition (Prechtel et al. 2001; Fowler et al. 2007). Studies in southeastern Canada and the northeastern US have documented that  $\text{SO}_4^{2-}$  concentrations in surface waters have also been decreasing in response to declining atmospheric deposition, but the surface water responses are often complicated by internal sources and sinks of S within ecosystems (Johnson and Mitchell 1998; Mitchell and Alewell 2007). The most important potential internal S sources are weathering of S minerals, mineralization of organic S, and desorption of adsorbed  $\text{SO}_4^{2-}$  (Fig. 2). These internal sources can contribute to  $\text{SO}_4^{2-}$  leaching to surface waters and may delay their recovery from acidification. The importance of such responses was emphasized in a series of studies (Church et al. 1989, 1992; Cosby et al. 1991) that demonstrated the importance of  $\text{SO}_4^{2-}$  adsorption/desorption characteristics of soils. Such results have been included in more extensive analyses of the factors affecting the  $\text{SO}_4^{2-}$  adsorption characteristics of soils (Harrison et al. 1989; Courchesne 1992) and how they influence both long-term (e.g., Rochelle et al. 1987) and short-term (e.g., Nodvin et al. 1986; Huntington et al. 1994;

Houle and Carignan 1995) soil  $\text{SO}_4^{2-}$  dynamics and fluxes in surface waters.

There has been considerable effort placed in the evaluation of S watershed budgets in North America including comparisons among S pools and S fluxes (e.g., Johnson and Mitchell 1998; Mitchell et al. 1992a, b; Jeffries et al. 2003a, b; Watmough et al. 2005; Mitchell and Alewell 2007). These syntheses have found that there is a wide range of atmospheric deposition of S among sites that is a direct function of spatial patterns of  $\text{SO}_2$  emissions. Also, the long-term temporal patterns of S deposition have been clearly linked to historical changes in  $\text{SO}_2$  emissions (Driscoll et al. 1998, 2001; Lynch et al. 2000; Butler et al. 2001; Likens et al. 2001, 2002). Sulfate losses in drainage waters exceed estimated atmospheric inputs for most studies that have calculated watershed S budgets in southeastern Canada (Houle and Carignan 1995; Houle et al. 1997; Beall et al. 2001; Eimers and Dillon 2002; Eimers and Houle 2005; Duchesne and Houle 2006) and the northeastern US (Likens and Bormann 1995; Driscoll et al. 1998; Likens et al. 2002; Park et al. 2003). The application of biogeochemical models for estimating watershed S dynamics have also suggested discrepancies as estimated by the difference between estimated atmospheric S inputs and drainage losses as  $\text{SO}_4^{2-}$  (Chen et al. 2004; Gbondo-Tugbawa et al. 2002). The role of discrepancies in ecosystem S budgets needs to be understood when making policy decisions relating the expected effect of changing emissions and resultant atmospheric deposition to ecosystem response including the response of surface waters (e.g., Driscoll et al. 2001; Sullivan and Cosby 2005). The contribution of substantial internal S sources can affect the establishment of critical and target loads for S deposition (Nilsson and Grennfelt 1988). A substantial internal S source may require setting target loads at a lower level compared to sites with little or no internal S contribution to surface waters. The application of critical loads has been extensively used in Europe (Hall et al. 2001) and is currently being applied throughout Canada and the U.S. (Dupont et al. 2005; Jeffries and Ouimet 2005; Porter et al. 2005; Ouimet et al. 2006; Burns et al. 2008; EPA 2009).

There are, however, substantial difficulties in calculating accurate S mass balances for watersheds. Relatively accurate estimates can be made for “wet



**Fig. 2** Major sulfur pools and fluxes in a forested watershed

only” (or bulk) atmospheric S inputs and drainage water losses (Likens and Bormann 1995; Mitchell et al. 2001b; Likens et al. 2002) although there can be considerable variability of wet and total deposition across landscapes (Ito et al. 2002; Weathers et al. 2000, 2006b). Comparisons of wet only and bulk S deposition estimates have generally found these measurements to be very similar in most studies (Shepard et al. 1989; Martin et al. 2000; Likens et al. 2002) although others have found some differences (Richter and Lindberg 1988; Staelens et al. 2005). Gaseous S emissions from plants and soil are not likely to play a substantial role in temperate forest S budgets although they may be potentially more important in tropical forests (Eaton et al. 1978; Haines et al. 1989; Mitchell et al. 1992a). There are major issues, however, regarding estimates of atmospheric S dry deposition, soil  $\text{SO}_4^{2-}$  adsorption/desorption, S mineral weathering and organic S immobilization/mineralization in watershed S cycling studies (Houle and Carignan 1992, 1995; Houle et al. 2001; Johnson and Mitchell 1998; Mitchell et al. 1992a, 2001a; Likens et al. 2002; Eimers et al. 2004a).

Dry deposition is a chemical and site specific process that depends on the characteristics of the receiving surface and the local meteorology (Weathers et al. 2006b). For studying national trends in deposition, the US operates the Clean Air Status and Trends Network (CASTNET; <http://www.epa.gov/castnet/index.html>) and Canada operates the Canadian Air and Precipitation Monitoring Network (CAPMoN; [http://www.msc.ec.gc.ca/capmon/index\\_e.cfm](http://www.msc.ec.gc.ca/capmon/index_e.cfm)). These networks use an inferential approach (Hicks et al. 1987, 1991) where atmospheric concentrations are measured and multiplied by a modeled deposition velocity to obtain the deposition flux. Each network uses a different deposition velocity model with CASTNET using the Multilayer Model (Meyers et al. 1998) and CAPMoN using the Routine Deposition Model (Brook et al. 1999; Zhang et al. 2009). Spatial interpolation of the values at the network sites is difficult as there is a need to account for the changes in the underlying surface (Clarke et al. 1997; Weathers et al. 2006b). Regional air quality models such as A Unified Regional Air-quality Modeling System (AURAMS) (Moran et al. 2008; Smyth et al. 2007, 2008) and the Community Multiscale Air Quality model (CMAQ; Byun and Schere

2006) provide the spatial estimates of deposition, but there are limited data from multi-year studies from these regional models. Direct deposition of cloud droplets to vegetation surfaces (cloud water deposition) can also contribute substantially to atmospheric S inputs at some sites, especially those at high elevations (Lovett et al. 1982; Lovett 1984; Weathers et al. 1995, 2000, 2006b; Baumgardner et al. 2003; Sickles and Grimm 2003). However, because of a lack of data and the limited area affected by the watersheds used in our study, we will not address this component of deposition in our paper.

Sulfate adsorption characteristics influence S budgets by regulating the adsorption and desorption of sulfate in soils. At the Hubbard Brook Experimental Forest in New Hampshire, temporal patterns in solute  $\text{SO}_4^{2-}$  concentrations following tree harvesting have been attributed to changes in  $\text{SO}_4^{2-}$  adsorption characteristics caused by nitrification induced acidification (Nodvin et al. 1986; Mitchell et al. 1989). The highest  $\text{SO}_4^{2-}$  adsorption potential is found in highly weathered soils with substantial concentrations of Fe and Al sesquioxides (Reuss and Johnson 1986; Johnson and Mitchell 1998). Highly weathered soils in eastern North America are predominantly found south of the limit of Pleistocene glaciation, where soil forming processes have been active for hundreds of thousands of years, versus the 10,000–15,000 year-old soils of our study sites that were affected by more recent Wisconsinian glaciations (Rochelle et al. 1987). Within our study area, the contribution of soil  $\text{SO}_4^{2-}$  adsorption/desorption in affecting  $\text{SO}_4^{2-}$  loss or retention over multiple years is relatively small because the size of the adsorbed sulphate pool is small compared to soils in unglaciated regions (Rochelle et al. 1987; Houle and Carignan 1995; Driscoll et al. 1998).

The precipitation and dissolution of  $\text{SO}_4^{2-}$  minerals do not generally dominate inorganic  $\text{SO}_4^{2-}$  dynamics (e.g., Johnson et al. 1982; Courchesne and Hendershot 1990; Faseth et al. 1991; Alewell et al. 1996; Lükewille et al. 1996) except for some sites in Europe with extremely high soil solution  $\text{SO}_4^{2-}$  and aluminum concentrations that may result in jurbanite formation (Nordstrom 1982; Khana et al. 1987). For many well-studied watersheds the results suggest that the weathering of S minerals is not the major source of S (Likens et al. 2002; Mitchell and Alewell 2007). However, in some watersheds there

may be substantial amounts of S minerals in some geological formations (e.g., shales, sulfidic micaeous phyllites) that may constitute an important internal source of S (Mitchell et al. 1986; Shanley et al. 2005). The contribution of the weathering of S minerals can affect the  $\text{SO}_4^{2-}$  isotopic composition of drainage waters in those watersheds where S minerals are important (Bailey et al. 2004).

For temperate forests generally more than 84% of S occurs within the mineral soil and generally more than 80% of this S is organic (e.g., carbon-bonded S, ester sulfates) (Houle and Carignan 1992; Mitchell et al. 1992a, b). The contribution of organic S dynamics to watershed budgets has been the focus of considerable attention due to its large pool size of which small changes could have dramatic effects on S export in drainage waters. Evidence using natural  $\text{SO}_4^{2-}$  isotope abundances (both  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$ ) from studies of forest ecosystems both in North America (Gélineau et al. 1989; Zhang et al. 1998; Alewell et al. 1999; Eimers et al. 2004c; Eimers and Houle 2005; Schiff et al. 2005) and Europe (Alewell and Gehre 1999; Mayer et al. 1995; Novák et al. 2005, 2007) has indicated the importance of the immobilization of  $\text{SO}_4^{2-}$  and organic S mineralization in affecting ecosystem S dynamics.

### New approaches for assessing watershed sulfur budgets

Given the decreases in S emissions and deposition that have been measured in eastern North America since the 1970s (Fig. 1), there is an important need to examine how these changes have affected the S export from watersheds throughout the region. Our overarching objective was to evaluate and compare S budgets in well-studied watersheds in the northeastern US and southeastern Canada. We were also able to compare changes in S budgets over the period of our study. In Table 1 we give information for each site including: the watershed's name, location, size, soil type, dominant vegetation, and key references. The locations of the watersheds are shown in Fig. 3. We have developed a series of calculations that use a consistent approach to provide common metrics for comparing similarities and differences among the watershed S budgets. Watersheds were selected based upon the availability of information on S

precipitation inputs and drainage water  $\text{SO}_4^{2-}$  losses over an extended period. Some of the watersheds have had additional studies on S pools and fluxes including the utilization of isotopic analyses. We have based our mass budget analyses on the well-established procedures and assumptions that have been used extensively for small watersheds (Likens and Bormann 1995; Moldan and Cerny 1995). As is typical for watershed studies that have well-documented hydrology and chemistry, we have substantial confidence in the estimates of S precipitation inputs and S drainage losses.

We hypothesized that most of the watersheds in our 15 study sites would show S outputs that exceed inputs due to losses through some combination of S mineral weathering, losses from soil organic matter, and desorption from the mineral soil. This hypothesis is supported by numerous studies that have found net S losses in drainage water in forested watersheds of southeast Canada and the northeast US. Our study approach recognized the difficulty and uncertainty of quantifying dry deposition inputs of S (e.g., Lovett 1994; Weathers et al. 2000, 2006b). Here, we applied two approaches to the entire study period for estimating dry S inputs using empirical equations using dry deposition estimates from (1) CASTNET and (2) CAPMoN. We also made comparisons using the regional CMAQ and AURAMS model for year 2002 (the year for which model results were available during our study period). All statistical analyses were performed using SAS for Windows® Version 9.1.3 (SAS Institute Inc., Cary, NC, USA).

### Precipitation and drainage water

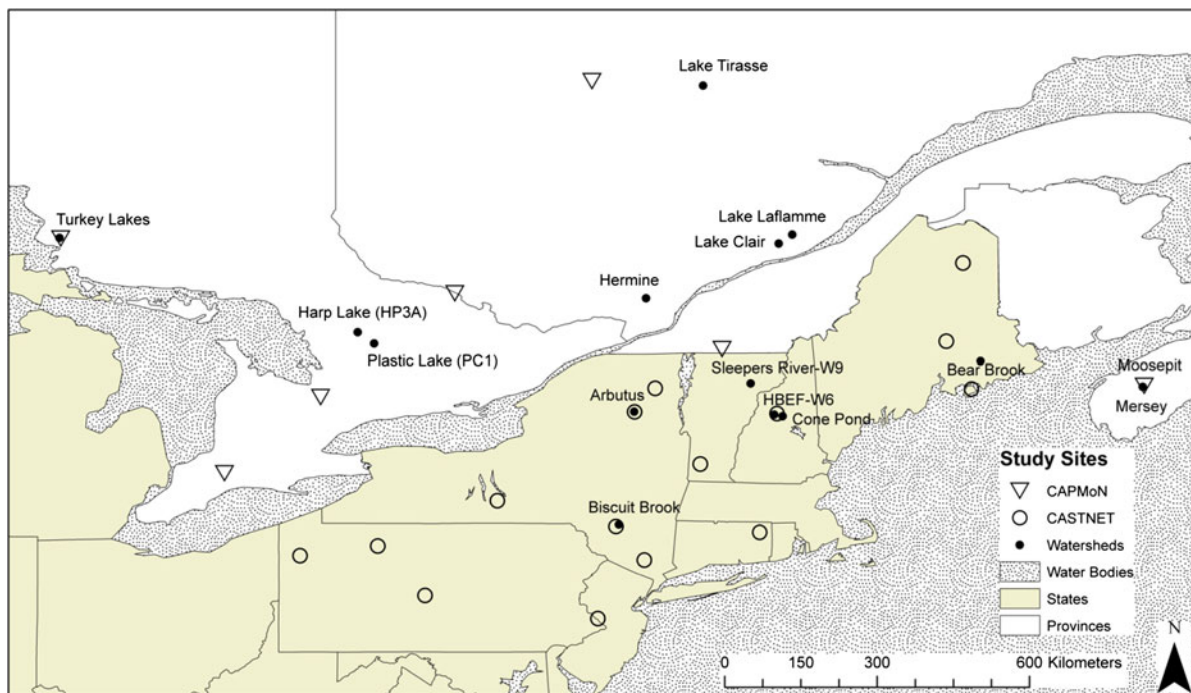
This synthesis is based on calendar-year budgets of atmospheric inputs and drainage water exports. The period of record and basic information on the frequency of collection and any differences among sites in measurement approaches (e.g., bulk deposition versus wet-only deposition) are provided in Table 2. Annual values for precipitation and drainage amounts were based upon the summation of measurements during each year for each site. Concentrations for precipitation and stream export were volume weighted (e.g., by precipitation amount or discharge rate) for each collection period, converted to fluxes using precipitation and drainage water values and summed for each calendar-year. For most watersheds,

**Table 1** Northeastern US and southeastern Canadian watersheds used in sulfur budget analyses

Watershed	Short name	Location	Latitude	Longitude	Size (ha)	Soil type	Dominant vegetation	Selected references
Biscuit Brook	Biscuit	Catskill Mountains, New York, USA	41.98	-74.5	959	Dystrudept and Fragiudept Inceptisols	<i>Acer rubrum</i> , <i>Fagus grandifolia</i> , <i>Acer saccharum</i> , <i>Betula Alleghaniensis</i>	Stoddard and Murdoch (1991), Murdoch and Shanley (2006)
Cone Pond	Cone	White Mountains, New Hampshire, USA	43.9	-71.6	33.4	Typic and Lithic Haplorthods	<i>Picea rubens</i> , <i>Abies balsamea</i> , <i>Tsuga canadensis</i> , <i>F. grandifolia</i> , <i>A. saccharum</i>	Bailey et al. (2004), Mitchell et al. (2008)
HBEF-W6	HBEF	White Mountains, New Hampshire, USA	43.93	-71.75	13.2	Typic and Lithic Haplorthods	<i>A. saccharum</i> , <i>F. grandifolia</i> , <i>B. alleghaniensis</i> , <i>A. balsamea</i> , <i>P. rubea</i>	Likens and Bormann (1995), Likens et al. (1990, 2002)
Arbutus	Arbutus	Adirondack Mountains, New York, USA	43.98	-74.23	352	Uplands: Becket-Mundal (coarse-loamy, mixed, frigid typic Haplorthods); Wetlands: Greenwood Mucky peats	<i>F. grandifolia</i> , <i>A. saccharum</i> , <i>T. canadensis</i>	Mitchell et al. (2001b, 2006)
Mersey Watershed	Mersey	Kejimikujik National Park, Nova Scotia, Canada	44.42	-65.23	29,500	Uplands are shallow podsols with gleysols and fens and peat bogs	<i>P. glauca mariana</i> , <i>A. balsamea</i> , <i>Pinus strobus</i> , <i>T. canadensis</i> , <i>A. saccharum</i> , <i>B. papyrifera</i> , <i>F. grandifolia</i>	Yanni et al. (2000)
Moosepit Watershed	Moosepit	Kejimikujik National Park, Nova Scotia, Canada	44.42	-65.23	1700	Upland are shallow podsols composed of granitic and quartzite materials. In areas of poor drainage, gleysols and fens and peat bogs	<i>P. glauca mariana</i> , <i>A. balsamea</i> , <i>P. strobus</i> , <i>T. canadensis</i> , <i>A. saccharum</i> , <i>B. papyrifera</i> , <i>Fagus grandifolia</i>	Yanni et al. (2000)
Sleepers River, Watershed 9	Sleepers	Vermont, USA	44.48	-72.17	40.5	Typic Dydrudepts (40%), Humic Dydrudepts (20%), Aquic Dystric Eutrudepts (20%), Typic Humaquepts (20%)	<i>A. saccharum</i> , <i>B. alleghaniensis</i> , <i>Fraxinus americana</i>	Shanley et al. (2005, 2008)
Bear Brook (East)	Bear	Maine, USA	44.87	-68.1	10.3	Typic Haplorthods	<i>F. grandifolia</i> , <i>A. rubrum</i> L., <i>A. saccharum</i> , <i>B. alleghaniensis</i> , <i>P. rubens</i>	David et al. (1990), Stam et al. (1992), Norton and Fernandez (1999)

Table 1 continued

Watershed	Short name	Location	Latitude	Longitude	Size (ha)	Soil type	Dominant vegetation	Selected references
Plastic Lake (PC1)	Plastic	Ontario, Canada	45.18	-78.83	23.3	Acidic podsols in uplands, organic soils and gleysols in depressions and wetlands	<i>Pinus strobus</i> , <i>Tsuga canadensis</i> , <i>Quercus rubra</i> , <i>Thuja occidentalis</i> , <i>Picea mariana</i>	Dillon and LaZerte (1992), LaZerte (1993)
Harp Lake (HP3A)	Harp	Ontario, Canada	45.38	-79.12	3.7	Acidic podsols in uplands, organic soils and gleysols in depressions and wetlands	<i>Acer saccharum</i> , <i>Acer rubrum</i> , <i>Populus grandidentata</i> , <i>Fagus grandifolia</i>	Eimers and Dillon (2002), Eimers et al. (2004b)
Hermine	Hermine	Quebec, Canada	45.98	-74.02	5.1	Sandy orthic or gleyed ferro-humic and humo-ferric podsols or Cryorthods	<i>A. saccharum</i> , <i>F. grandifolia</i> , <i>B. alleghaniensis</i>	Biron et al. (1999), Bélanger et al. (2002), Courchesne et al. (2005)
Lake Clair	Clair	Centre récréotouristique de Duchesnay, Quebec, Canada	46.95	-71.67	226	Entic Haploorthod or Ferro-Humic Podzol	<i>Acer saccharum</i> , <i>Fagus grandifolia</i> , <i>Betula alleghaniensis</i>	Houle et al. (1997)
Lake Laflamme	Laflamme	Réserve faunique des Laurentides, Quebec, Canada	47.3281	-71.126944	68	Typic Haploorthods or Orthic Humo-Ferric Podzol	<i>Abies balsamea</i> , <i>Picea glauca</i> , <i>Betula papyrifera</i>	Houle and Carignan (1992, 1995), Houle et al. (2001)
Turkey Lakes (average of watersheds 31-35, 37-39, 42, 46-47, 49-50)	Turkey	Ontario, Canada	47.05	-84.4	3-115	Uplands: Haplohumods and Haploorthods; Wetland: Haplohemists	<i>A. saccharum</i> (90%), <i>B. alleghaniensis</i> , <i>P. strobus</i> , <i>P. glauca</i>	Beall et al. (2001), Jeffries et al. (1988)
Lake Tirasse	Tirasse	Réserve faunique Ashuapmushuan, Quebec, Canada	49.75	-73.01	56	Typic Haploorthods or Orthic Humo-Ferric Podzol	<i>P. mariana</i> , <i>P. banksiana</i>	Duchesne and Houle (2006)



**Fig. 3** Location of watersheds in the northeastern U.S. and southeastern Canada with information on sulfur budgets and location of CASTNET and CAPMoN sites used in developing Eq. 1

precipitation amounts were measured daily and concentration measurements were based upon weekly collections. Discharge measurements were made at different intervals (ranging from continuous to mean average daily values) and chemistry measurements were generally made on a weekly basis with any exceptions noted. For Turkey Lakes the discharge and stream export data used here were an average of 13 watersheds (watersheds 31–35, 37–39, 42, 46–47, 49–50). We selected the 17-year period from 1985 to 2002 as the period for comparison because of data completeness. For some watersheds only a portion of this period was available. Also, for a few watersheds longer records were available beyond our selected study period. We compared S budgets among these watersheds using different periods and did not find any marked differences in overall budget patterns.

#### Dry deposition

Obtaining accurate and consistent long-term estimates of dry deposition for use in mass balance studies is a major challenge. Many of the watersheds used in this comparison did not have dry deposition

estimates available on site; therefore, we required an approach for estimating annual dry deposition for all sites that (1) provided estimates for each site, (2) was consistent in methodology across sites, and (3) was able to make annual estimates over time as S emissions changed. To meet these criteria, we developed a two-step approach. In the first step, we used  $\text{SO}_2$  concentration data from dry deposition monitoring networks in the northeastern US (CASTNET) and southeastern Canada (CAPMoN) along with annual eastern US  $\text{SO}_2$  emissions data to develop an empirical relationship. This relationship was then used to determine the spatial and temporal patterns of  $\text{SO}_2$  concentrations in the air at the watershed locations. To match the geographical area of the watersheds we were studying, we used  $\text{SO}_2$  concentration data only from sites  $>40^\circ\text{N}$  latitude and  $<85^\circ\text{W}$  longitude from both the CASTNET (US) and CAPMoN (Canada) dry deposition monitoring networks. There were 23 sites included in this geographical area, 16 in the US and seven in Canada (Fig. 3). The estimates of annual  $\text{SO}_2$  emissions were derived from the US EPA estimates of the Acid Rain Program sources located in eastern US (EPA Regions

**Table 2** Watershed data utilized in sulfur budget calculations (1985–2002)

Watershed	Period of record for each site within study period	Precipitation collection method (bulk or wet only)	Discharge measurements intervals	Discharge chemistry measurement intervals
Biscuit Brook	1985–2002	Wet only (NADP)	15 min	Weekly + Events
Cone Pond	1985–2002	Bulk	Continuous	Weekly+
HBEF	1985–2002	Bulk	Continuous	Weekly
Arbutus	1985–2002	Wet only (NADP)	15 min	Weekly
Mersey	1985–2002	Wet only	Continuous	Weekly
Moosepit	1985–2002	Wet only	Continuous	Weekly
Sleepers River	1992–2002	Bulk	5 min	Weekly + Events
Bear Brook	1985–2001	Wet only	5 min	Weekly
Plastic Lake	1985–1997	Bulk	10 min	Weekly + Events
Harp Lake	1985–1997	Bulk	10 min	Weekly + Events
Hermine	1995–1997, 2001–2002	Bulk	Daily	Daily
Lake Clair	1985–2002	Bulk	Read every 5 s/1 h average recorded	Weekly
Lake Laflamme	1985–1996, 1999–2002	Wet Only	Read every 5 s/1 h average recorded	Weekly
Turkey Lakes <sup>a</sup>	1985–2002	Wet only	Continuous	Daily during peak melt, bi-weekly remainder of year
Lake Tirasse	1997–2002	Bulk	Read every 5 s/1 h average recorded	Weekly

<sup>a</sup> Mean values of Discharge for watersheds 31, 32, 33, 34, 35, 37, 38, 39, 42, 46, 47, 49, 50

1 through 5): <http://camddataandmaps.epa.gov/gdm/index.cfm?fuseaction=emissions.wizard>. Previous analyses have indicated that the Canadian SO<sub>2</sub> emissions constitute a small (~10%) proportion of the total and hence would not markedly affect this relationship (for further information see: Canada-United States Air Quality Agreement: 2008 Progress Report; Vet and Ro 2008). The yearly estimated eastern US SO<sub>2</sub> emissions declined substantially for the period of our study (1985–2002) (Fig. 1), illustrating the importance of including a temporal component to the dry deposition calculations. Note that SO<sub>2</sub> emissions data were available from the US EPA only for the years 1985, 1990, 1995, and then annually thereafter, so for the periods 1986–1989 and 1991–1994, we linearly interpolated emissions data based on emissions before and after the period for 1987, 1988, 1992 and 1993 for calculations for those years.

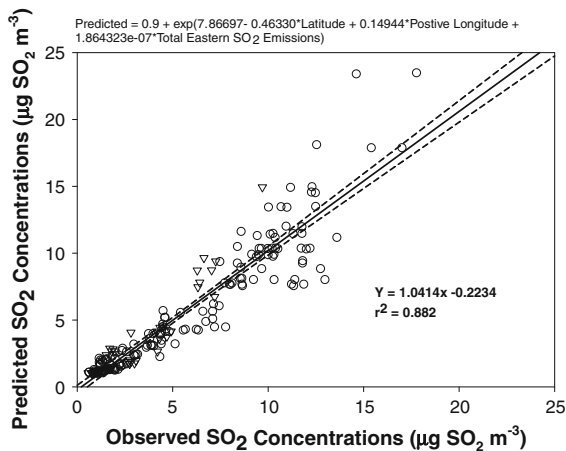
A variety of possible relationships between annual average atmospheric SO<sub>2</sub> concentration and latitude, longitude and annual eastern US SO<sub>2</sub> emissions were

tested and we found that an asymptotic exponential function (Eq. 1, Fig. 4) explained 88% of the variation in predicted versus observed SO<sub>2</sub> concentrations with little bias (slope of 1.0).

$$\text{SO}_{2\text{ave}} = 0.9 + \exp(7.867 - 0.4633 * \text{lat} + 0.1494 * \text{long} + 1.863e - 7 * \text{toteastSO}_2) \quad (1)$$

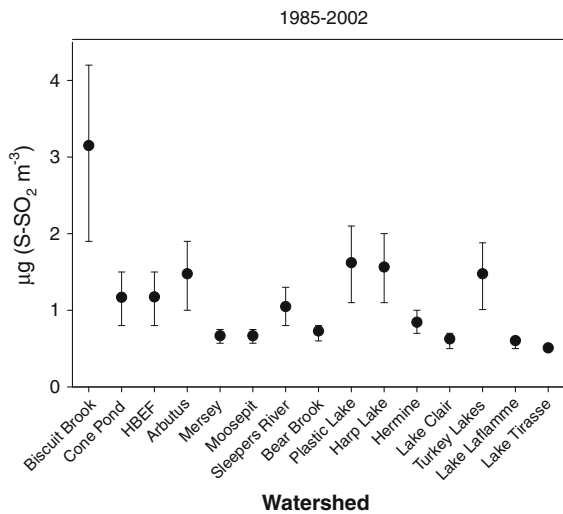
where SO<sub>2ave</sub> is the annual average SO<sub>2</sub> concentration (µg m<sup>-3</sup>) for each site and year, lat is ° north latitude, long is ° west longitude, and toteastSO<sub>2</sub> is the total emissions (tons per year) of SO<sub>2</sub> from the eastern US for the year.

Using Eq. 1, the average annual SO<sub>2</sub> concentrations for each site were calculated. The mean and range of SO<sub>2</sub> concentrations across the years of the study are shown in Fig. 5 with sites ranked by latitude from south to north. For all sites the mean SO<sub>2</sub> concentration was 1.2 µg S-SO<sub>2</sub> m<sup>-3</sup>, the maximum concentration was at Biscuit Brook (3.2 µg S-SO<sub>2</sub> m<sup>-3</sup>) and the minimum concentration at Lake Tirasse (0.5 µg S-SO<sub>2</sub> m<sup>-3</sup>)



**Fig. 4** Solid line is regression equation of observed versus predicted annual SO<sub>2</sub> concentrations as a function of latitude, longitude and SO<sub>2</sub> emissions from the eastern United States for CASTNET and CAPMoN sites. Dashed lines are 95% confidence limits. Sites used in these calculation are shown in Fig. 3

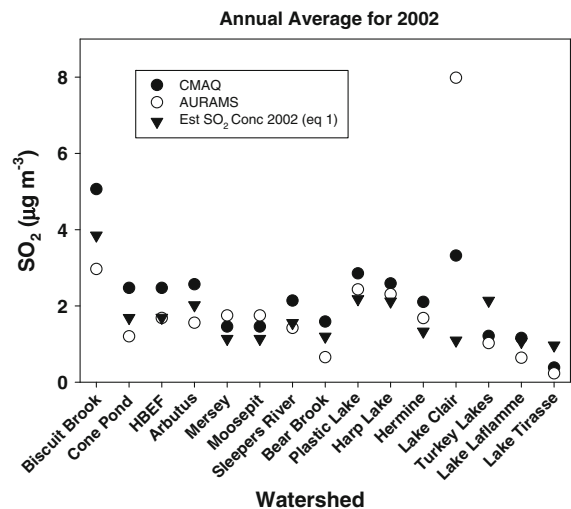
(Fig. 5). To examine the potential impact of local sources and the effects of not including chemical transformations, we compared output for the year 2002 (the only year where complete data were available) from the regional air quality models CMAQ (Byun and Schere 2006) and AURAMS (Smyth et al. 2008), which were run using the full US and Canada emissions inventories, with the values predicted for each watershed from Eq. 1. Details on specific CMAQ and



**Fig. 5** Range and mean values of calculated SO<sub>2</sub> concentrations for the period from 1985 to 2002 using formulation based upon latitude, longitude and eastern U.S. SO<sub>2</sub> emissions for the study sites (see text for details)

AURAMS applications can be found in Appel and Roselle (2009) and Moran et al. (2008), respectively. Together the results from the CMAQ and AURAMS models give a sense of the results expected from a spatially explicit treatment of emissions, transport, and transformation compared with the simplified empirical approach used in our study. There is a notable difference between the CMAQ and AURAMS estimates for the Lake Clair watershed where a local SO<sub>2</sub> source (Alcoa aluminum smelter) was located within the AURAMS grid cell (Moran, unpublished data), but not within the CMAQ grid cell. Figure 6 shows that concentrations estimated using Eq. 1 are quite similar to those predicted by the two regional air quality models for most sites. If we exclude the Lake Clair SO<sub>2</sub> estimate for AURAMS, there are strong and highly significant correlations ( $r = 0.731$  to  $0.886$ ,  $p < 0.01$ ) among SO<sub>2</sub> concentration estimates for each site using Eq. 1, AURAMS and CMAQ.

In the second step, we used data from all 86 CASTNET sites for years 1990–2003 and from the 11 CAPMoN sites for years 1998–2002 (the number of years with results varied among sites) to derive an empirical relationship between annual average SO<sub>2</sub> concentration and modeled total S dry deposition (SO<sub>2</sub> plus SO<sub>4</sub> particles) using the CASTNET and CAPMoN data sets separately. We also evaluated the CASTNET relationship using a subset of sites (e.g., forested sites in the northeast US) and found no substantial difference in this empirical relationship using this more



**Fig. 6** Predicted SO<sub>2</sub> concentrations for study region for CMAQ model, AURAMS model, and Eq. 1 for year 2002

limited data set. Previous studies comparing concentration values from CASTNET and CAPMoN at a co-located site in Egbert, showed that the measured air concentrations for  $\text{SO}_2$  and  $\text{SO}_4^{2-}$  were almost identical for the CASTNET and CAPMoN measurements, but the CAPMoN deposition estimates for  $\text{SO}_2$  and  $\text{SO}_4^{2-}$  were substantially greater due to different models of deposition velocity (Vet et al. 2005). As a result of the differences in the deposition models, the data from each network give different relationships between  $\text{SO}_2$  concentration and total S dry deposition (Fig. 7). The two equations are:

CASTNET Equation:

$$\text{kg S ha}^{-1} \text{ year}^{-1} = \text{SO}_2 (\mu\text{g S m}^{-3}) * 1.169 + 0.00572$$

$$r^2 = 0.901 \quad (2)$$

where  $\text{SO}_2$  values are average yearly concentrations for each CASTNET site.

CAPMoN Equation:

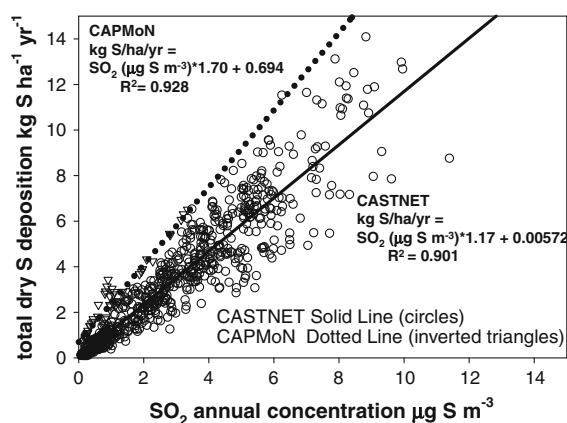
$$\text{kg S ha}^{-1} \text{ year}^{-1} = \text{SO}_2 (\mu\text{g S m}^{-3}) * 1.696 + 0.694$$

$$r^2 = 0.928 \quad (3)$$

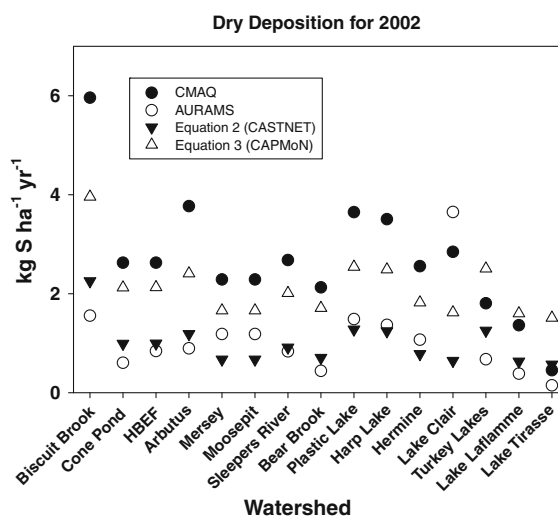
where  $\text{SO}_2$  values are average yearly concentrations for each CAPMoN site.

Equations 2 and 3 demonstrate the uncertainty associated with models of deposition velocity. Use of other deposition formulations such as those used in CMAQ and AURAMS will result in different

estimates of the deposition velocity. The various deposition models used in CASTNET, CAPMoN, CMAQ and AURAMS produce estimates of deposition velocity that are comparable to field estimates (e.g., Meyers et al. 1998; Finkelstein et al. 2000; Pleim et al. 2001; Zhang et al. 2001), but are not in agreement with each other. Comparing the estimates from Eqs. 2 and 3 with the deposition estimates from CMAQ and AURAMS provides a more complete depiction of the uncertainty associated with the deposition velocity estimates at the watershed sites than just Eqs. 2 and 3 alone (Fig. 8). As noted previously there is a notable difference for the Lake Clair watershed where a local  $\text{SO}_2$  source is located within the AURAMS grid cell (Moran, unpublished data), but not within the CMAQ grid cell, and not sufficiently close to the network monitoring sites to influence the regression equation. There was a strong and significant correlation ( $r = 0.607\text{--}0.880$ ,  $p \leq 0.02$ ) between the dry S deposition estimates among sites for year 2002 using AURAMS, CMAQ, Eqs. 2 and 3 excluding Lake Clair. Because data are needed for the complete period of the study, for the remainder of our current analyses we have used Eqs. 2 and 3 to estimate dry deposition at each watershed location and to provide a sense of how the uncertainty in the dry deposition estimates affects S mass balance budgets.



**Fig. 7** Calculated relationships between annual  $\text{SO}_2$  concentrations and total dry deposition of S using data for CASTNET (Equation 2) and CAPMoN (Equation 3) sites



**Fig. 8** Predicted S dry deposition for study region for CMAQ model, AURAMS model, Equation 2 (CASTNET data) and Equation 3 (CAPMoN data) for year 2002

## Watershed budgets

### Water budgets

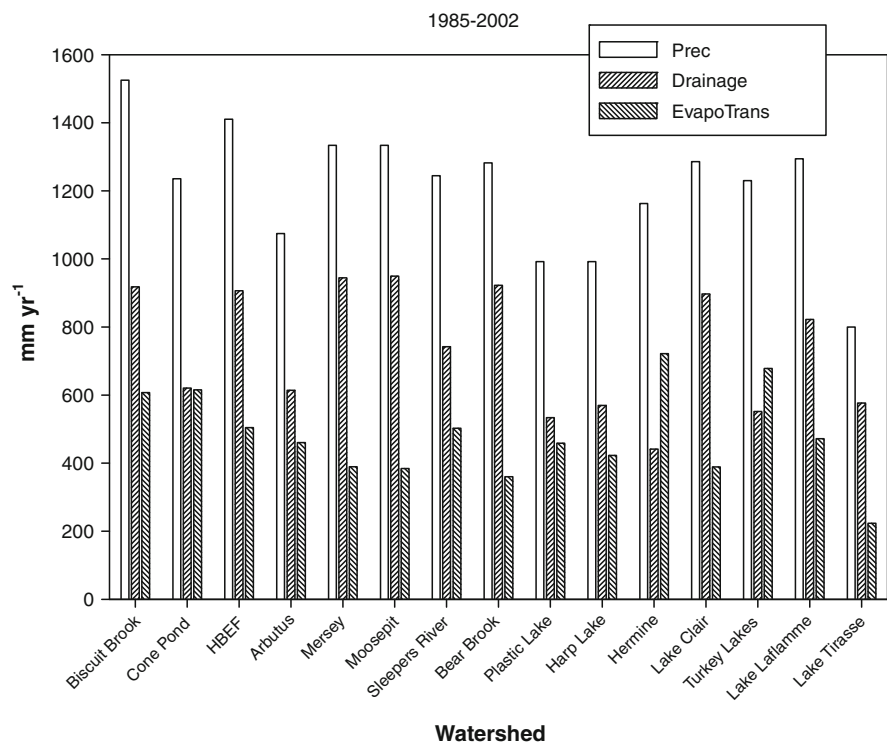
The mean annual precipitation for all the watersheds for the period from 1985 to 2002 was 1215 mm with minimum annual precipitation (800 mm) occurring at Lake Tirassee and the maximum (1800 mm) at Biscuit Brook (Fig. 9). The average annual discharge amount for this same period for all watersheds was 737 mm with minimum (441 mm) at Hermine and a maximum (969 mm) at Moosepit. Average annual calculated evapotranspiration (precipitation–discharge) for this same period for all watersheds was 478 mm with minimum (223 mm) at Lake Tirassee and a maximum (721 mm) at Hermine. The mean annual precipitation, discharge and evapotranspiration values for each watershed are provided in Fig. 9.

Sulfur budgets including estimates of dry deposition using formulations based upon measurements from CASTNET and CAPMoN

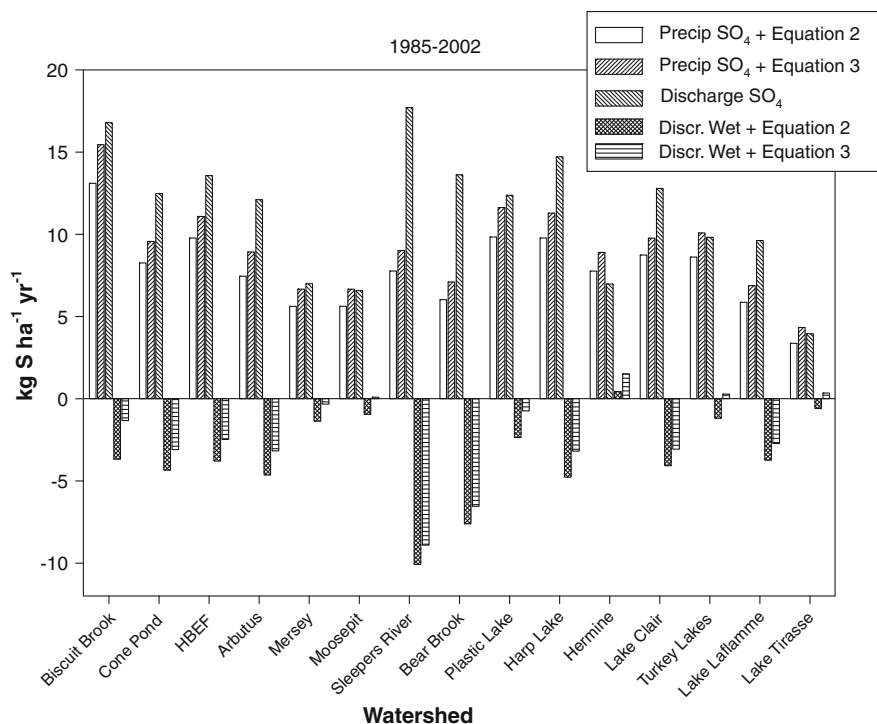
The use of the CASTNET-based Equation 2 estimates for total dry S deposition resulted in a mean dry

deposition of  $1.4 \text{ kg S ha}^{-1} \text{ year}^{-1}$  with a minimum at Lake Tirassee ( $0.6 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ) and a maximum at Biscuit Brook ( $3.7 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ). The use of CAPMoN-based Equation 3 resulted in substantially higher mean dry deposition estimates of  $2.7 \text{ kg S ha}^{-1} \text{ year}^{-1}$  with a minimum at Lake Tirassee ( $1.6 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ) and a maximum at Biscuit Brook ( $6.0 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ). Average  $\text{SO}_4^{2-}$  discharge among watersheds was  $11.3 \text{ kg S ha}^{-1} \text{ year}^{-1}$  with a large range among watersheds from  $4.0 \text{ kg S ha}^{-1} \text{ year}^{-1}$  (Lake Tirassee) to  $17.7 \text{ kg S ha}^{-1} \text{ year}^{-1}$  (Sleepers River) (Fig. 10). The inclusion of the estimates of S dry deposition combined with S precipitation inputs were used to calculate average differences in total atmospheric inputs versus losses from discharge for each of the watersheds (Fig. 10). The use of Eq. 2 resulted in a mean discrepancy of  $-3.5 \text{ kg S ha}^{-1} \text{ year}^{-1}$  with the smallest ( $+0.4 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ) at Hermine, which was the only watershed to show net S retention using this estimate of total atmospheric deposition, and a largest ( $-10.0 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ) at Sleepers River. Using this latter estimate, four watersheds out of the 15 sites showed net S retention (Hermine:  $+1.5 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ; Lake Tirassee:  $+0.3 \text{ kg S ha}^{-1} \text{ year}^{-1}$ ; Turkey Lakes:

**Fig. 9** Mean annual precipitation, discharge values and calculated evapotranspiration (precipitation–discharge) for each watershed for the period from 1985 through 2002. For details on data utilized see Table 1



**Fig. 10** Annual fluxes of total atmospheric sulfur deposition (precipitation combined with estimates of dry deposition based upon Eqs. 2 or 3, see text for details) with discrepancy calculated by subtracting sulfur in discharge from the estimates of total atmospheric sulfur deposition



+0.3 kg S ha<sup>-1</sup> year<sup>-1</sup> Moosepit: +0.1 kg S ha<sup>-1</sup> year<sup>-1</sup>). The use of Eq. 3 resulted in a lower mean discrepancy of -2.2 kg S ha<sup>-1</sup> year<sup>-1</sup> with the smallest difference (-0.2 and +0.2 kg S ha<sup>-1</sup> year<sup>-1</sup>) at (Mersey and Moosepit, respectively) and largest difference (-10.1 kg S ha<sup>-1</sup> year<sup>-1</sup>) at Sleepers River.

### Individual watershed sulfur budget results and comparisons among watersheds

#### Temporal patterns

For the study period from 1985 through 2002, regression analyses ( $p \leq 0.05$ ) were used to evaluate temporal changes in various components of the S budget of each site. In addition to the flux measurements, we also estimated annual, volume-weighted sulfate concentrations of precipitation and stream discharge in  $\mu\text{mol SO}_4^{2-} \text{ l}^{-1}$ . Using appropriate conversions from mass to molar values and dividing the annual  $\text{SO}_4^{2-}$  flux value of precipitation or discharge ( $\text{kg S-SO}_4^{2-} \text{ ha}^{-1} \text{ year}^{-1}$ ) by water flux ( $\text{mm year}^{-1}$ ), we converted these values to  $\mu\text{mol SO}_4^{2-} \text{ l}^{-1}$ . Using linear regression, we found that the three measured attributes that showed

the most consistent and significant relationships over time were S precipitation flux and  $\text{SO}_4^{2-}$  concentration in precipitation and drainage water (Table 3). For the 15 watershed locations two pairs of sites used the same S precipitation inputs (Plastic Lake and Harp Lake; Mersey and Moosepit) and each pair showed significant decreases in S inputs. For the other 11 watershed locations, seven showed a significant reduction in precipitation S inputs over time. Similar, but fewer significant results were also found for changes in  $\text{SO}_4^{2-}$  concentration in precipitation with the two paired sites showing declines. For the other 11 locations, five sites showed significant ( $p \leq 0.05$ ) decreases in  $\text{SO}_4^{2-}$  concentration in precipitation over time. As expected, there were also significant changes in  $\text{SO}_4^{2-}$  concentration in drainage water with eight of 15 sites showing significant declines in  $\text{SO}_4^{2-}$  concentration over time (Table 3). The lack of significant changes over time in  $\text{SO}_4^{2-}$  concentration for some sites was likely due in part to the absence of complete coverage within the period of analyses for those sites with lowest deposition and fewest years of record (see Table 2 for more details on years covered).

An examination of the patterns of S precipitation inputs, discharge and discrepancies in annual values showed considerable variation among years. Annual

**Table 3** Temporal changes (annual values) for precipitation and discharge from 1985 through 2002 (significant values  $p \leq 0.05$  are given in **bold**)

Watershed	Precip. flux (kg S ha <sup>-1</sup> year <sup>-1</sup> )	<i>p</i> value	Precip. conc. ( $\mu\text{mol SO}_4^{2-}$ l <sup>-1</sup> year <sup>-1</sup> )	<i>p</i> value	Discharge conc. ( $\mu\text{mol SO}_4^{2-}$ l <sup>-1</sup> year <sup>-1</sup> )	<i>p</i> value
Biscuit Brook	<b>-0.28</b>	<b>0.0003</b>	<b>-0.61</b>	<b>0.0001</b>	<b>-1.29</b>	<b>&lt;0.0001</b>
Cone Pond	<b>-0.16</b>	<b>0.0276</b>	-0.19	0.3201	-0.60	0.1401
HBEF	<b>-0.17</b>	<b>0.0130</b>	<b>-0.39</b>	<b>0.0002</b>	<b>-0.81</b>	<b>&lt;0.0001</b>
Arbutus	<b>-0.18</b>	<b>0.0002</b>	<b>-0.58</b>	<b>&lt;0.0001</b>	<b>-0.86</b>	<b>0.0142</b>
Mersey Watershed	<b>-0.14</b>	<b>0.0011</b>	<b>-0.35</b>	<b>0.0015</b>	0.08	0.481
Moosepit Watershed	<b>-0.14</b>	<b>0.0011</b>	<b>-0.35</b>	<b>0.0015</b>	0.05	0.6398
Sleepers River	0.06	0.6357	-0.04	0.8768	-0.02	0.9734
Bear Brook	-0.07	0.3161	-0.09	0.4945	-1.24	0.2489
Plastic Lake	<b>-0.22</b>	<b>0.0031</b>	<b>-0.68</b>	<b>0.0007</b>	<b>-3.45</b>	<b>0.0557</b>
Harp Lake	<b>-0.22</b>	<b>0.0031</b>	<b>-0.68</b>	<b>0.0007</b>	<b>-2.08</b>	<b>0.0083</b>
Hermine	-0.05	0.5519	0.09	0.6294	3.00	0.3528
Lake Clair	<b>-0.24</b>	<b>0.0423</b>	-0.34	0.1546	<b>-0.79</b>	<b>&lt;0.0001</b>
Lake Laflamme	<b>-0.12</b>	<b>0.0347</b>	-0.21	0.0771	<b>-1.18</b>	<b>&lt;0.0001</b>
Turkey Lakes	<b>-0.21</b>	<b>0.0004</b>	<b>-0.41</b>	<b>0.0004</b>	<b>-0.72</b>	<b>0.0084</b>
Lake Tirasse	-0.07	0.5748	-0.27	0.5162	<b>-0.59</b>	<b>0.0078</b>

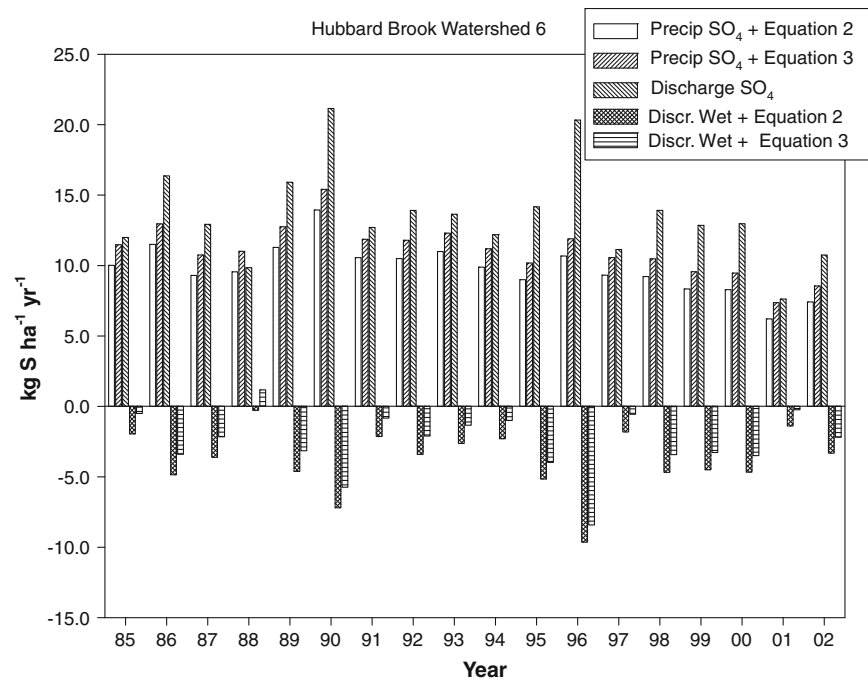
SO<sub>4</sub><sup>2-</sup> budgets are provided for Hubbard Brook Experimental Forest Watershed 6 (Fig. 11) and Turkey Lakes (Fig. 12) along with their respective annual hydrological fluxes (Figs. 13, 14, respectively) to illustrate the importance of this annual variation. For example, note the correspondence with high water discharge and SO<sub>4</sub><sup>2-</sup> loss at the Hubbard Brook Experimental Forest in 1990 and 1996. At Turkey Lakes the highest discharge and SO<sub>4</sub><sup>2-</sup> loss occurred in 1988. Similar results are apparent for all the watersheds used in our study (detailed results not shown). The importance of watershed hydrology in affecting temporal variation of SO<sub>4</sub><sup>2-</sup> concentrations has been reported previously within the study region with particular attention to the role of drying and wetting in affecting SO<sub>4</sub><sup>2-</sup> mobilization (Eimers and Dillon 2002; Jeffries et al. 2003b; Eimers et al. 2007; Mitchell et al. 2008).

#### Comparisons of sulfur budget differences among watersheds

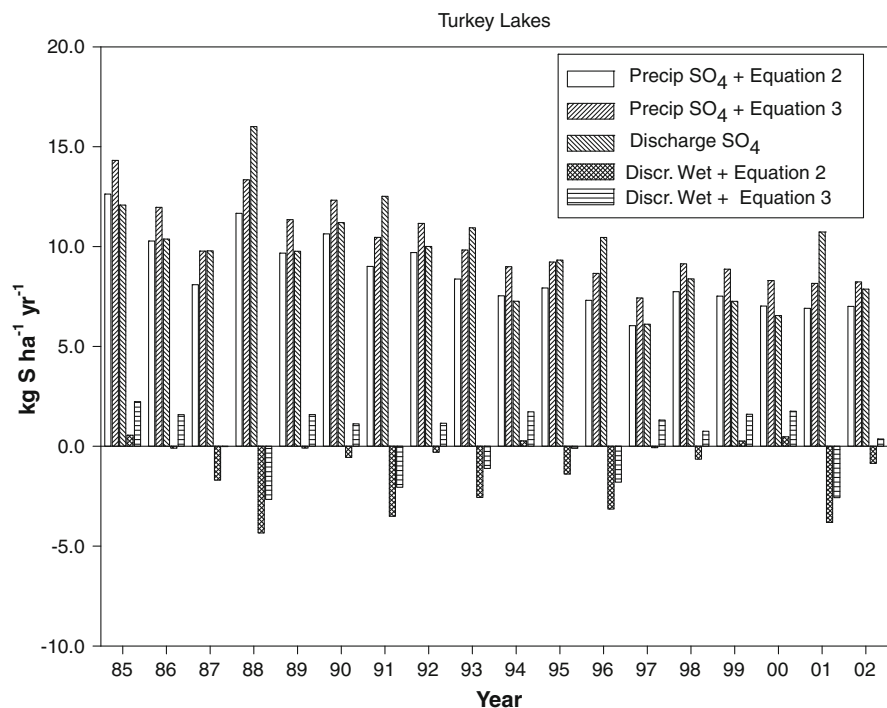
To facilitate the temporal and spatial comparisons of the watersheds, the discrepancies in annual S budgets in kg S ha<sup>-1</sup> year<sup>-1</sup> were converted to mean annual, volume-weighted concentration values ( $\mu\text{mol SO}_4^{2-}$  l<sup>-1</sup>) using the annual stream discharge for each

watershed. This conversion to concentrations facilitates comparisons of S budget discrepancies among watersheds and over time since the interannual S drainage water fluxes for each catchment are greatly affected by differences in annual water discharge as discussed above. The mean annual concentration discrepancies (ranked from most negative to most positive) for each site using precipitation plus dry deposition (Equation 2: CASTNET values; Equation 3: CAPMoN values) are provided in Fig. 15. Only Lake Laflamme showed a significant trend in concentration discrepancies over time (total deposition using Eq. 2: slope = 0.908  $\mu\text{mol SO}_4^{2-}$  year<sup>-1</sup>,  $p \leq 0.0001$ ; total deposition using Eq. 3: slope = 0.880  $\mu\text{mol SO}_4^{2-}$  year<sup>-1</sup>,  $p = 0.0001$ ) suggesting that these discrepancies have remained relatively similar over the period of study for almost all of the watersheds. Typical results of annual variation are shown for Arbutus, Biscuit Brook and Moosepit watersheds (Fig. 16). Some of the variation in these concentration discrepancies among years can be attributed to watershed precipitation and hence hydrological relationships. This relationship was evaluated by regressing the annual concentration discrepancies against annual precipitation for each site. For eight out of the fifteen sites there was a significant inverse relationship between annual

**Fig. 11** Annual sulfur budgets for Hubbard Brook Watershed 6: bulk precipitation + either Equation 2 or Equation 3 dry deposition estimates, discharge and discrepancy using Eq. 2 or Eq. 3 deposition estimates



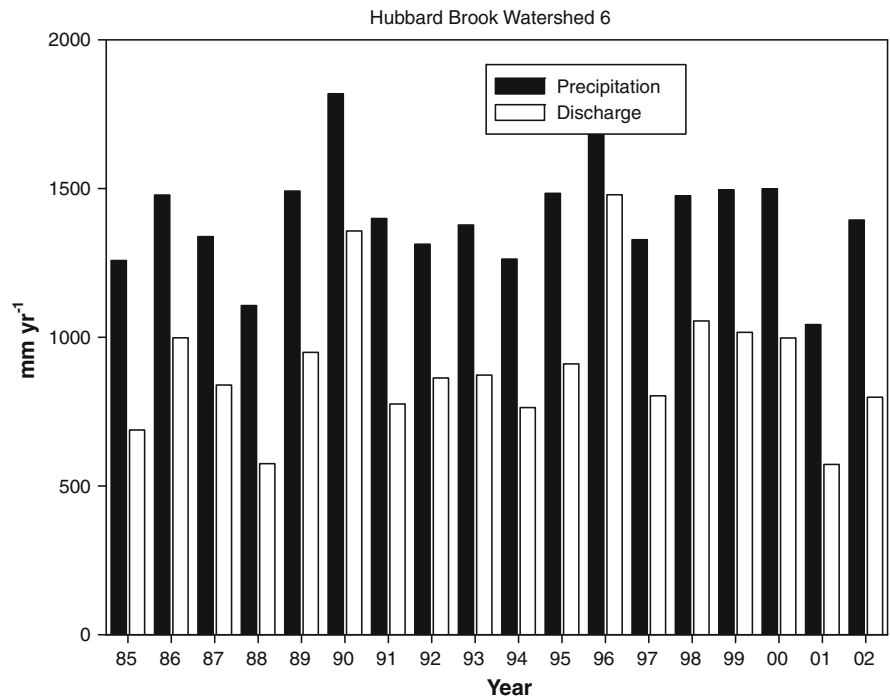
**Fig. 12** Annual sulfur budgets for Turkey Lakes watersheds: bulk precipitation + either Equation 2 or Equation 3 dry deposition estimates, discharge (average of watersheds as detailed in text) and discrepancy using either Eq. 2 or Eq. 3 deposition estimates



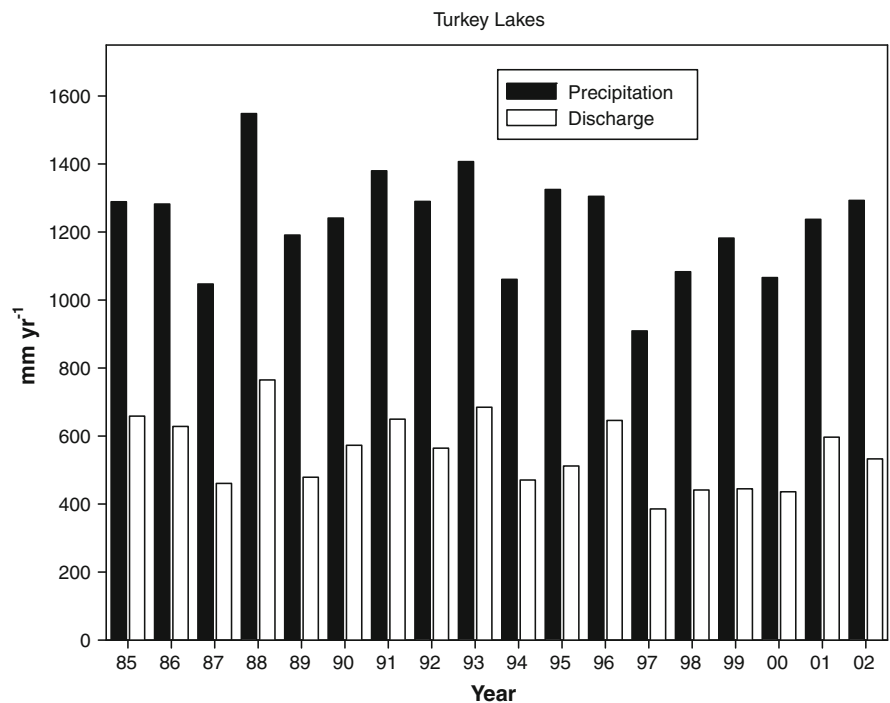
precipitation and concentration discrepancies (Table 4) suggesting the importance of hydrology in the regulation of SO<sub>4</sub><sup>2-</sup> in drainage waters.

Studies at many of these sites (Plastic Lake: Eimers and Dillon 2002; Eimers et al. 2004a, c; Arbutus: Mitchell et al. 2006, 2008; Sleepers River: Mitchell

**Fig. 13** Annual hydrology (precipitation and discharge) for Hubbard Brook Watershed 6



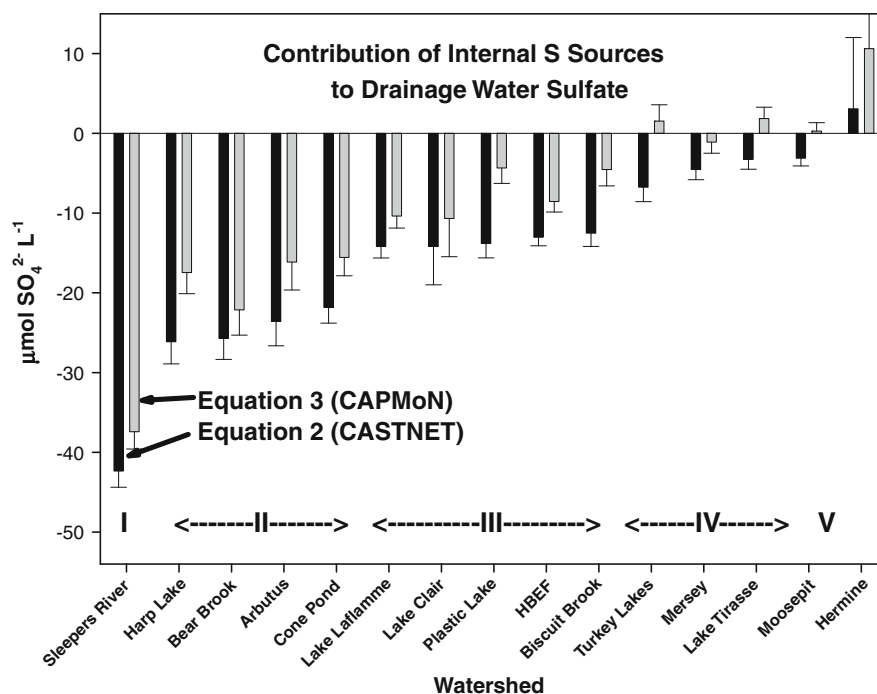
**Fig. 14** Annual hydrology (precipitation and discharge) for Turkey Lakes (discharge averaged for watersheds as detailed in text)



et al. 2008, Shanley et al. 2008; Cone Pond: Mitchell et al. 2008; Turkey Lakes: Schiff et al. 2005) have established the importance of changing hydrology in

affecting S dynamics due to the linkages among watershed wetness, redox conditions and  $\text{SO}_4^{2-}$  mobilization.

**Fig. 15** Classification of watersheds based upon the  $\text{SO}_4^{2-}$  concentration contributed by internal sulfur sources. Error bars are 1 standard error. See text for details on rationale of classification



An examination of the mean  $\text{SO}_4^{2-}$  concentration discrepancies suggest that the 15 sites can be categorized based upon the range of these values (Fig. 15). For more details on the S budgets and biogeochemistry of individual sites see Appendix I—Supplementary material.

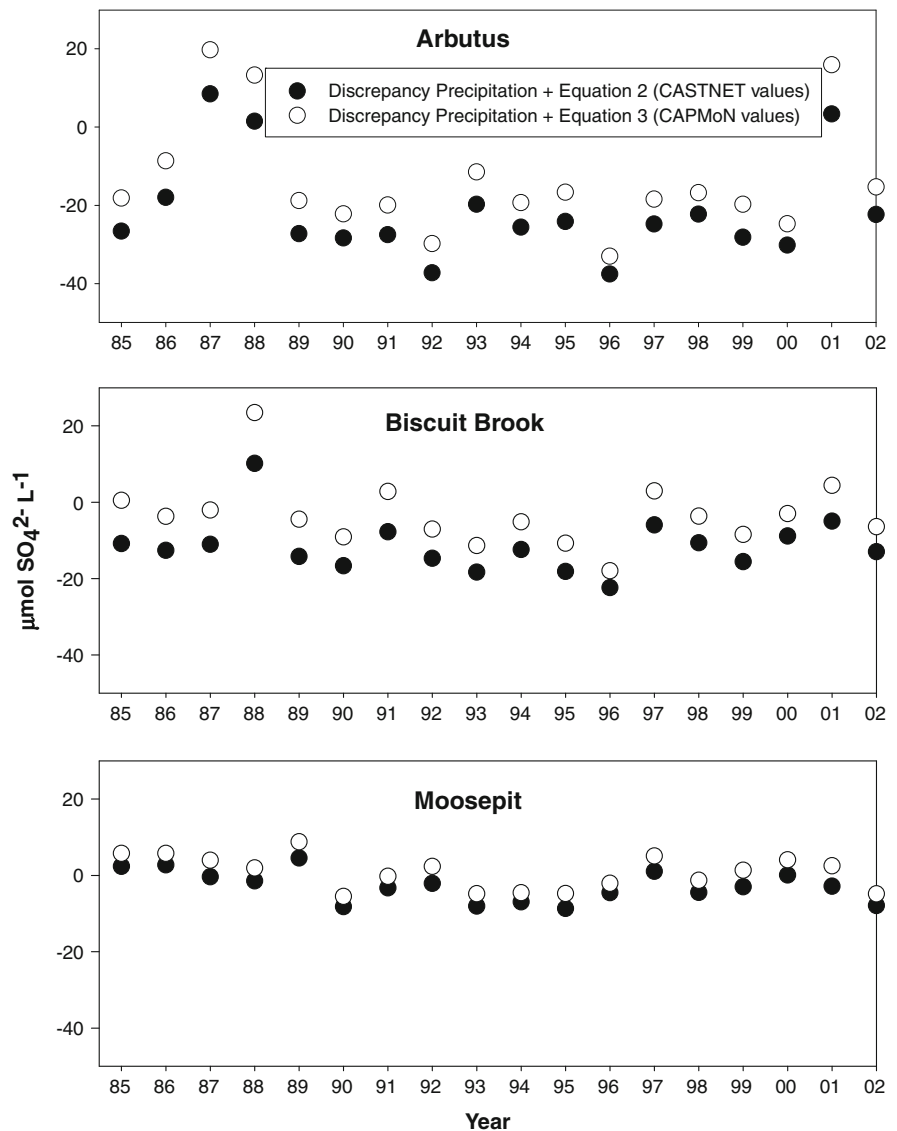
**Category I:** Clearly, Sleepers River has the greatest discrepancy ( $-37$  to  $-42 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$ ) of all other sites by  $>13 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$  and hence the largest internal S source of all watersheds in the current study. The source of this discrepancy has been previously identified, using mass budgets and stable isotopic analyses of S sources, to be weatherable S minerals (Bailey et al. 2004; Shanley et al. 2005), with possible occasional contributions from reoxidized secondary sulfides (Shanley et al. 2008).

**Category II:** These sites (Bear Brook, Harp, Arbutus, and Cone Pond) have concentration discrepancies from  $-20$  to  $-27 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$  (using Eq. 2 estimates) or  $-14$  to  $-23 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$  (using Eq. 3 estimates) and some have evidence of internal S sources. At Bear Brook experimental work using “mineral soil bags” has shown that much of the short-term variation and response to S additions were due to changes in adsorbed  $\text{SO}_4^{2-}$  (David et al. 1990). Previous isotopic analyses ( $\delta^{34}\text{S-SO}_4^{2-}$ ) at

Bear Brook suggested that most of the  $\text{SO}_4^{2-}$  in discharge can be attributed to S derived from atmospheric deposition, but an additional unknown internal S source was likely present (Stam et al. 1992). Harp Lake Watershed (HP3A) is predominantly upland, and as a consequence  $\text{SO}_4^{2-}$  concentrations in stream water are much less variable over time (Eimers et al. 2008; Seip et al. 1985) than nearby Plastic Lake as discussed below. Previous analysis at Arbutus watershed, including estimates of dry deposition, suggested that an internal S source was required to balance the S budget (Park et al. 2003). Results using  $\text{SO}_4^{2-}$  isotopic evidence ( $\delta^{18}\text{O}$  and  $\delta^{34}\text{S}$ ) (Campbell et al. 2006) and spatial patterns of  $\text{SO}_4^{2-}$  concentrations (Piatek et al. 2009) have suggested that some subcatchments of the Arbutus Watershed have a strong internal S source. Previous work at Cone Pond has suggested the potential importance of a fire in 1820 that heavily burned 85% of the watershed. It has been suggested that this fire reduced the organic matter content with a resultant effect on  $\text{SO}_4^{2-}$  dynamics (Mitchell et al. 2008).

**Category III:** These sites (Lake Laflamme, Plastic, Lake Clair, HBEF and Biscuit Brook) have concentration discrepancies lower (i.e., less negative) than Category II. The values range from  $-12$  to  $-14 \mu\text{mol}$

**Fig. 16** Mass balance annual discrepancy value in  $\text{SO}_4^{2-}$  concentrations calculated from the differences in discharge–precipitation + dry deposition from Equation 2 (CASTNET values) (closed circle) and precipitation + dry deposition from Equation 3 (CAPMoN values) (open circle) for Arbutus, Biscuit Brook and Moosepit. See text for further details on the calculations



$\text{SO}_4^{2-} \text{ l}^{-1}$  (using Eq. 2 estimates) or  $-3$  to  $-10 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$  (using Eq. 3 estimates). For this category, although the discrepancies are smaller than those in Category II, they all show a net loss of  $\text{SO}_4^{2-}$ . Previous studies at Lake Laflamme Watershed have indicated that soil  $\text{SO}_4^{2-}$  sorption should adjust rapidly (within 4 years) to changing S loads but that desorption alone cannot explain long-term net  $\text{SO}_4^{2-}$  losses (Houle and Carignan 1995). An oxygen isotope study of the dissolved  $\text{SO}_4^{2-}$  in soil solution demonstrated that 32–61% of the  $\text{SO}_4^{2-}$  leaving the catchment had interacted with organic S in the soil (Gélineau et al. 1989). Previous work at Plastic Lake

(PC1) Watershed has indicated that sulfate export was strongly influenced by the presence of a large conifer-*Sphagnum* swamp (e.g., LaZerte 1993; Eimers and Dillon 2002; Eimers et al. 2007; Aherne et al. 2008). Stable isotopic analyses have shown that changes in  $\text{SO}_4^{2-}$  concentration at this watershed are associated with microbial redox processes (Eimers et al. 2004a, b). Previous investigations at Lake Clair Watershed attributed net S losses to a combination of  $\text{SO}_4^{2-}$  desorption and/or organic S mineralization (Houle et al. 1997). At the Hubbard Brook Experimental Forest, there has been considerable effort associated with the evaluation of S budgets since 1964 (e.g.,

**Table 4** Discrepancies calculated using  $\mu\text{mol SO}_4^{2-}$  concentrations and the volume of discharge for, precip + Equation 2 or precip + Equation 3 ( $p \leq 0.05$ )

Site	Slope annual precip (mm) versus conc. discrepancy (precip. + Equation 2)	<i>p</i> value	Slope annual precip (mm) versus conc. discrepancy (precip. + Equation 3)	<i>p</i> value
Arbutus	-0.065	0.008	-0.078	0.005
Biscuit Brook	-0.016	0.015	-0.021	0.006
Cone Pond	-0.028	0.007	-0.037	0.002
Harp	-0.066	0.013	-0.089	0.003
HBEF	-0.019	<0.0001	-0.024	<0.0001
Mersey	-0.015	0.037	-0.019	0.017
Moosepit	-0.012	0.027	-0.016	0.007
Turkey Lakes	-0.029	0.019	-0.039	0.004

Likens and Bormann 1995; Likens et al. 2002). This previous work has also suggested that there is a discrepancy in the net hydrologic S budgets (precipitation inputs minus streamwater outputs) for the various watersheds of the Hubbard Brook Experimental Forest including W6 used in the current study. The use of isotopic analyses ( $\delta^{34}\text{S}$ ) of  $\text{SO}_4^{2-}$  has suggested that the discrepancy is likely due to the mineralization of a small fraction of the large organic S pool (Alewell et al. 1999). The use of the PnET-BGC simulation model also support the importance of the mineralization of the organic S pool in the soil as the major contributor to the discrepancy in the net hydrologic S budget (Gbondo-Tugbawa et al. 2002). There is no evidence that weathering is a substantial S source (Likens et al. 2002; Bailey et al. 2004). At Biscuit Brook, the results of the current study indicate a substantial net watershed loss of S and differ from an earlier study in which S inputs were estimated to approximately balance outputs when dry S deposition was assumed to equal 33% of wet S deposition (Stoddard and Murdoch 1991). This earlier study was only for 2 years and used results from a period (1984–1985) with higher rates of S deposition than for the average for the entire period (1985–2002) of the current study.

**Category IV:** These sites (Turkey Lakes, Mersey, Lake Tirassee and Moosepit) are characterized by concentration discrepancies near zero ( $-6$  to  $+3 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$ ). For all of these sites, except Mersey, the utilization of the higher dry deposition estimates using Eq. 3 resulted in small net S retention. All four of the sites in Category IV are in Canada and relatively remote from major sources of anthropogenic S deposition

(Fig. 3). Previous studies at Turkey Lakes have ascribed some of the decrease in stream  $\text{SO}_4^{2-}$  to losses of exchangeable  $\text{SO}_4^{2-}$  from upper soils layers (Morrison et al. 1992; Beall et al. 2001; Morrison and Foster 2001). Watershed S discrepancies at Lake Tirassee Watershed of  $0.9 \text{ kg ha}^{-1} \text{ year}^{-1}$  have been previously documented using throughfall S deposition, plus the contribution of dissolved organic sulfur (DOS) in incoming precipitation as a surrogate for total S deposition from 1997–2003. These previous studies have also suggested that mineralization of soil organic S was the likely source of the excess S (Duchesne and Houle 2006). For Moosepit Watershed, the results of the current study provide a different estimate from that of Yanni et al. (2000) who ascribed a discrepancy between measured wet CAPMoN S deposition and export to unmeasured fog deposition.

**Category V:** The Hermine site is very distinctive with exceptionally positive discrepancies ( $>+14 \mu\text{mol SO}_4^{2-} \text{ l}^{-1}$ ) suggesting strong S retention in this watershed. However, Hermine has a discharge ratio (discharge/precipitation) of about 38%, a low value for forested watersheds of Northeastern North America. Along with the Turkey Lakes and Lake Tirassee catchments (using Eq. 3), Hermine is the only watershed apparently retaining S on a mean annual basis using Eqs. 2 and 3. In all cases, these years were much dryer than average with less than 1150 mm precipitation and high summer temperatures that caused streamflow to cease for prolonged periods during the growing season. Sulfur was lost ( $-1.2$  to  $-4.2 \text{ kg S ha}^{-1} \text{ year}^{-1}$  with Eq. 3) from the watershed when the Hermine experienced cooler and much wetter conditions.

## Conclusions

Our analyses underscore the importance of having accurate estimates of atmospheric deposition, both wet and dry, for assessing S budgets for watersheds in southeastern Canada and the northeastern US. Although relatively accurate values for the wet deposition of S are available, the contribution of dry deposition is more difficult to quantify because of the inherent difficulties in estimating dry deposition to the landscape. An evaluation of the contribution of dry deposition is needed, however, for making both temporal and spatial analyses of S budgets and relating these budgets to changes in the emissions of SO<sub>2</sub>. We developed an empirical approach for estimating dry deposition based upon results available from both the US (CASTNET) and Canadian (CAPMoN) dry deposition networks to provide a range of estimates. This empirical approach included the use of regional S emissions to the general region, SO<sub>2</sub> concentrations, and spatial location. We also compared these empirical results with the results from the CMAQ and AURAMS models for 2002 to help further our understanding of approaches for evaluating dry deposition. Using estimates of total S deposition we evaluated S budgets for 15 watershed sites and found substantial differences among these budgets.

For some watersheds such as Sleepers River (Category I) the importance of the weathering of S-bearing minerals resulted in a substantial source of SO<sub>4</sub><sup>2-</sup> to drainage waters. For other relatively remote sites in Canada such as Turkey Lakes, Mersey, Lake Tirassee and Moosepit (Category IV), the results suggested that S inputs are relatively closely matched with SO<sub>4</sub><sup>2-</sup> losses in drainage waters. For the majority of the watersheds included in our study (especially categories I to III), there was an additional source of S and our analyses suggest that this contribution is relatively constant within a site among years. A range of the contribution of this internal S source can be calculated by using a range of SO<sub>4</sub><sup>2-</sup> discrepancy values (i.e., -3 to -27 μmol SO<sub>4</sub><sup>2-</sup> l<sup>-1</sup>; Fig. 15; Categories II and III) and using the average annual discharge for the 15 sites (737 mm year<sup>-1</sup>). We excluded Sleepers River with its known mineral S sources and those watersheds (Category IV) that have not been subjected to lower levels of atmospheric deposition of S. This calculation suggests that internal S sources can contribute an additional 1 to

6 kg S ha<sup>-1</sup> year<sup>-1</sup> in the annual SO<sub>4</sub><sup>2-</sup> export. Other studies have suggested that although there may be some contribution from the desorption of SO<sub>4</sub><sup>2-</sup> that accumulated during periods of higher S deposition, the SO<sub>4</sub><sup>2-</sup> adsorption capacities of the soils in this region are relatively small (Rochelle et al. 1987; Mitchell et al. 1992a, b; Houle and Carignan 1995). Mass balance calculations suggest that only a relatively small proportion of the discrepancy in watershed S budgets can be attributed to SO<sub>4</sub><sup>2-</sup> desorption (Driscoll et al. 1998). Although the source of this additional S is not known with complete certainty, stable isotopic analyses studies both in North America (Alewell et al. 1999, 2000; Likens et al. 2002; G lineau et al. 1989) and Europe (Nov k et al. 2005, 2007) have strongly suggested that the mineralization of organic S in soil could be an important source. The results of the current study suggest that this internal source is more important

for those sites that have been subjected to elevated levels of atmospheric S deposition (e.g., Harp Lake, Arbutus, Plastic, HBEF, Biscuit Brook) compared to those sites that have had relatively low amounts of atmospheric S inputs (Mersey, Lake Tirassee, Moosepit). This finding is supported by a significant ( $r^2 = 0.54$ ,  $p = 0.02$ ) positive relationship between the S budget discrepancy (including dry deposition estimates provided by Eq. 2) versus S precipitation inputs during the period of the study for watersheds in Categories III and IV for which S weathering inputs are likely very small based upon their known mineralogy.

The factors that affect the mobilization of previously stored S are not completely understood although we do know that watershed hydrology can have a major influence on SO<sub>4</sub><sup>2-</sup> drainage losses and retention. Some of these factors are associated with effects related to the mobilization of SO<sub>4</sub><sup>2-</sup> formed from previously reduced S especially in wetlands (e.g., Schiff et al. 2005; Eimers et al. 2007; Mitchell et al. 2008). Watershed wetness also affects the connectivity of surface waters to solute sources (Creed and Band 1998; Inamdar et al. 2004) and increased wetness would likely result in the enhanced SO<sub>4</sub><sup>2-</sup> movement from soils to streams. Changing climate conditions and the resultant effects on the hydrology and S biogeochemistry of eastern North American forests are likely to be important in the future (e.g., Campbell et al. 2009; Huntington et al. 2009).

The surface waters of southeastern Canada and the northeastern U.S. have shown substantial recovery from high levels of S deposition, but the contribution of these internal additional sources of S needs to be considered in evaluating the long-term recovery of terrestrial ecosystems. As atmospheric S inputs decrease, the relative importance of internal S sources in contributing to the  $\text{SO}_4^{2-}$  in drainage waters will increase. Moreover, recent evidence has suggested that soils within this region have become depleted in nutrient cations such as calcium and magnesium (Fernandez et al. 2003; Bailey et al. 2005; Sullivan et al. 2006). The atmospheric input of S and the mobilization of the  $\text{SO}_4^{2-}$  anion from internal S sources will not only exacerbate the loss of base cations but will also contribute to the continued acidification of soils and surface waters by the mobilization of  $\text{H}^+$  and toxic monomeric Al. Hence these internal S sources need to be taken into account not only in understanding watershed S biogeochemistry, but also in applying various management tools such as critical loads (Sverdrup and de Vries 1994; Dupont et al. 2005; Ouimet et al. 2006) for predicting the effects of changing atmospheric deposition on ecosystem responses.

**Acknowledgments** This work was supported by The Andrew W. Mellon Foundation, Environment Canada, Fonds de Recherche sur la nature et les technologies (FQRNT) du Québec, National Resources Canada, Natural Sciences and Engineering Research Council (NSERC) of Canada, US National Science Foundation including the LTER and LTREB programs, New York City Department of Environmental Protection, New York State Energy Research Development Authority, Northeast Ecosystem Research Cooperative (NERC), Ontario Ministry of Environment, US Forest Services, US EPA, and US Geological Survey. We acknowledge funding support for sample collection and analysis by the U.S. Environmental Protection Agency's Long-Term Monitoring program. The comments of Tom Huntington and J. E. Sickles on this manuscript were most helpful. Kim McEathron helped with some of the figures. The United States Environmental Protection Agency through its Office of Research and Development collaborated in the research described here. It has been subjected to Agency review and approved for publication.

## References

- Aherne JM, Futter MN, Dillon PJ (2008) The impacts of future climate change and sulphur emission reductions on acidification recovery at Plastic Lake, Ontario. *Hydrol Earth Syst Sci* 12:383–392
- Alewell C, Gehre M (1999) Patterns of stable S isotopes in a forested catchment as indicators for biological S turnover. *Biogeochemistry* 47:319–333
- Alewell C, Manderscheid B, Lükewille A et al (1996) Describing  $\text{SO}_4^{2-}$  dynamics in the Solling roof project with two different modelling approaches. *Water Air Soil Pollut* 85:1806–1810
- Alewell C, Mitchell M, Likens GE et al (1999) Sources of stream sulfate at the Hubbard Brook Experimental Forest: long-term analysis using stable isotopes. *Biogeochemistry* 44:281–299
- Alewell C, Mitchell M, Likens GE et al (2000) Assessing the origin of sulfate deposition at the Hubbard Brook Experimental Forest. *J Environ Qual* 29:759–767
- Appel KW, Roselle SJ (2009) Evaluation of CMAQ v4.7 sulfate predictions for 2002–2006. In: 8th annual community modeling and analysis (CMAS) conference, Chapel Hill, NC
- Bailey SW, Mayer B, Mitchell MJ (2004) Evidence for the influence of mineral weathering on stream water sulphate in Vermont and New Hampshire. *Hydrol Proc* 18:1639–1653
- Bailey SW, Horsley SB, Long RP (2005) Thirty years of change in forest soils of the Allegheny Plateau, Pennsylvania. *Soil Sci Soc Am J* 69:681–690
- Baumgardner RE, Lavery TF, Rogers CM et al (2002) Estimates of the atmospheric deposition of sulfur and nitrogen species: Clean Air Status and Trends Network, 1990–2000. *Environ Sci Technol* 36:2614–2629
- Baumgardner RE, Isil SS, Lavery TF et al (2003) Estimates of cloud water deposition at mountain acid deposition program sites in the Appalachian Mountains. *J Air Waste Manage Assoc* 53:291–308
- Beall FD, Semkin RG, Jeffries DS (2001) Trends in the output of first-order basins at Turkey Lakes Watershed, 1982–96. *Ecosystems* 4:514–526
- Bélanger N, Courchesne F, Côté B et al (2002) Simulation of soil chemistry and nutrient availability in a forested ecosystem of southern Quebec. Part II. Application of the SAFE model. *Environ Model Softw* 17:447–465
- Biron P, Roy AG, Courchesne F et al (1999) The effects of antecedent moisture conditions on the relationship of hydrology to hydrochemistry in a small forested watershed. *Hydrol Proc* 13:1541–1555
- Brook JR, Zhang L, Li Y et al (1999) Description and evaluation of a model of deposition velocities for routine estimates of dry deposition over North America. Part II: review of past measurements and model results. *Atmos Environ* 33:5053–5070
- Burns DA, Blett T, Haeuber R, Pardo LH (2008) Critical loads as a policy tool for protecting ecosystems from the effects of air pollutants. *Front Ecol Environ* 6:156–159
- Butler TJ, Likens GE, Stunder BJB (2001) Regional-scale impacts of Phase I of the Clean Air Act Amendments in the USA: the relation between emissions and concentrations, both wet and dry. *Atmos Environ* 35:1015–1028
- Byun DW, Schere KL (2006) Review of the governing equations, computational algorithms, and other components of the models-3 community multiscale air quality (CMAQ) modeling system. *Applied mechanics reviews*. *Am Soc Mech Eng* 59(2):51–77

- Campbell JL, Mitchell MJ, Mayer B (2006) Isotopic assessment of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  mobility during winter in two adjacent watersheds in the Adirondack Mountains, New York. *J Geophys Res* 111:G04007. doi:10.1029/2006JG000208
- Campbell JL, Rustad LE, Boyer EW et al (2009) Consequences of climate change for biogeochemical cycling in forests of northeastern North America. *Can J For Res* 39:264–284
- Canada-United States Air Quality Agreement: 1992 Progress report. ISBN:0-662-19759-3, 79 pp. [http://www.ec.gc.ca/pdb/can\\_us/qual/1992/english/index\\_e.html](http://www.ec.gc.ca/pdb/can_us/qual/1992/english/index_e.html)
- Canada-United States Air Quality Agreement: 2008 Progress report. ISSN 1487-1033, ISBN: 978-1-100-10516-1, 72 pp. [http://www.ec.gc.ca/cleanair-airpur/83930AC3-60EE-4A14-A241-A10A83D3B055/US\\_Canada\\_2008\\_En\\_web.pdf](http://www.ec.gc.ca/cleanair-airpur/83930AC3-60EE-4A14-A241-A10A83D3B055/US_Canada_2008_En_web.pdf)
- Chen L, Driscoll CT, Gbondo-Tugbawa S et al (2004) The application of an integrated biogeochemistry model (PnET-BGC) to five forested watersheds in the Adirondack and Catskill Regions of New York. *Hydrol Proc* 18:2631–2650
- Church MR, Thornton KW, Shaffer PW et al (1989) Direct/delayed response project: future effects of long term sulfur deposition on surface water chemistry in the northeastern and southern Blue Ridge province. In: EPA/600/3-89/061. U.S. Environmental Protection Agency, Washington, DC
- Church MR, Shaffer PW, Thornton KW et al (1992) Direct/delayed response project: future effects of long term sulfur deposition on stream chemistry in the mid-Appalachian region of the eastern United States. In: EPA/600/R-92/186. U.S. Environmental Protection Agency, Washington, DC
- Clair TA, Dillon PJ, Ion J et al (1995) Regional precipitation and surface water chemistry trends in southeastern Canada (1983–1991). *Can J Fish Aquat Sci* 52(1):197–212
- Clarke JF, Edgerton ES, Martin BE (1997) Dry deposition calculations for the CASTNET. *Atmos Environ* 31:3667–3678
- Cosby BJ, Ryan PF, Webb JR (1991) Mountains of western Virginia. In: Charles DF et al (eds) *Acidic deposition and aquatic ecosystems: regional case studies*. Springer-Verlag, New York
- Courchesne F (1992) Relationships between soil chemical properties and sulfate sorption kinetics in Podzolic soils from Quebec. *Can J Soil Sci* 72:467–480
- Courchesne F, Hendershot WH (1990) The role of basic aluminum sulfate minerals in controlling sulfate retention in acid forest soils in Quebec. *Soil Sci* 150:571–578
- Courchesne F, Côté B, Fyles J et al (2005) Recent changes in soil chemistry in a forested ecosystem of southern Québec, Canada. *Soil Sci Soc Am J* 69:1298–1313
- Creed IF, Band LE (1998) Export of nitrogen from catchments within a temperate forest: evidence for a unifying mechanism regulated by variable source area dynamics. *Water Resour Res* 34:3079–3093
- David MB, Fuller RD, Fernandez IJ et al (1990) Spodosol variability and assessment of response to acidification. *Soil Sci Soc Am J* 54:541–548
- Dennis I, Clair T, Driscoll C et al (2005) Distribution patterns of mercury in lakes and rivers of Northeastern North America. *Ecotoxicology* 14:113–123
- Dillon PJ, LaZerte BD (1992) Response of the Plastic Lake catchment, Ontario, to reduced sulphur deposition. *Environ Pollut* 77:211–217
- Driscoll CT, Postek KM (1996) The chemistry of aluminum in surface waters. In: Sposito G (ed) *The environmental chemistry of aluminum*. CRC, Boca Raton, pp 363–418
- Driscoll CT, Likens GE, Church MR (1998) Recovery of surface waters in the northeastern U.S. from decreases in atmospheric deposition of sulfur. *Water Air Soil Pollut* 105:319–329
- Driscoll CT, Lawrence G, Bulger A et al (2001) Acidic deposition in the northeastern US: sources, inputs, ecosystem effects, and management strategies. *BioScience* 51:180–198
- Driscoll CT, Driscoll KM, Roy KM et al (2003) Chemical response of lakes in the Adirondack region to declines in acidic deposition. *ES&T* 37:2036–2042
- Duchesne L, Houle D (2006) Base cation cycling in a pristine watershed of the Canadian boreal forest. *Biogeochemistry* 78:195–216
- Dupont J, Clair TA, Gagnon C et al (2005) Estimation of critical loads of acidity for lakes in northeastern United States and eastern Canada. *Environ Monit Assess* 109:275–291
- Eaton JS, Likens GE, Bormann FH (1978) The input of gaseous and particulate sulfur to a forest ecosystem. *Tellus* 30:546–551
- Eimers MC, Dillon PJ (2002) Climate effects on sulphate export from forested catchments in south-central Ontario, Canada. *Biogeochemistry* 61:355–367
- Eimers C, Houle D (2005) Sources, transport and fate of S in Canadian watersheds. In: Houle D (ed) *Effects on forests and soils. Chapter 5 Canadian acid deposition science assessment, 2004*. Environment Canada, Ottawa, pp 168–173
- Eimers MC, Dillon PJ, Schiff SL (2004a) A S isotope approach to determine the relative contribution of redox processes to net  $\text{SO}_4$  export from upland and wetland dominated catchments. *Geochim Cosmochim Acta* 68:3665–3674
- Eimers MC, Dillon PJ, Schiff SL (2004b) Sulphate flux from an upland forested catchment in south-central Ontario, Canada. *Water Air Soil Pollut* 152:3–21
- Eimers MC, Dillon PJ, Watmough SA (2004c) Long-term (18-year) changes in sulphate concentrations in two Ontario headwater lakes and their inflows in response to decreasing deposition and climate variations. *Hydrol Proc* 18:2617–2630
- Eimers MC, Watmough SA, Buttle JM et al (2007) Drought-induced sulphate release from a wetland in south-central Ontario. *Environ Monit Assess* 127:399–407
- Eimers MC, Buttle JM, Watmough SA (2008) Influence of seasonal changes in runoff and extreme events on dissolved organic carbon trends in wetland- and upland-draining streams. *Can J Fish Aquat Sci* 65:796–808
- EPA (2009) Risk and exposure assessment for review of the secondary national ambient air quality standards for oxides of nitrogen and oxides of sulfur. EPA-452/R-09-008a
- Faseth WJ, David MB, Vance GF (1991) Sulfate retention and cation leaching of forest soils in response to acid additions. *Can J For Res* 21:32–41
- Fernandez IJ, Rustad LE, Norton SA et al (2003) Experimental acidification causes soil base cation depletion in a New

- England forested watershed. *Soil Sci Soc Am J* 67: 1909–1919
- Finkelstein PL, Ellestad TG, Clarke JF et al (2000) Ozone and sulfur dioxide dry deposition to forests: observations and model evaluation. *J Geophys Res Atmos* 105:15365–15377
- Fowler D, Smith R, Muller J et al (2007) Long term trends in sulphur and nitrogen deposition in Europe and the cause of non-linearities. *Water Air Soil Pollut* 7:41–47
- Gauci V, Dise NB, Howell G et al (2008) Suppression of rice methane emission by sulfate deposition in simulated acid rain. *J Geophys Res* 113:G00A07. doi:10.1029/2007JG000501
- Gbondo-Tugbawa SS, Driscoll CT, Mitchell MJ et al (2002) A model to simulate the response of a northern hardwood forest ecosystem to changes in S deposition. *Ecol Appl* 12:8–23
- Gélineau M, Carignan R, Tessier A (1989) Study of the transit of sulfate in a Canadian Shield lake watershed with stable oxygen isotope ratios. *Appl Geochem* 4:195–201
- Gilmour CC, Henry EA, Mitchell R (1992) Sulfate stimulation of mercury methylation in freshwater sediments. *ES&T* 26:2281–2287. doi:10.1021/es00035a029
- Haines B, Black M, Bayer C (1989) Sulfur emissions from roots of the rain forest tree *Stryphnodendron excelsum*. In: Saltzman ES, Cooper WJ (eds) Biogenic sulfur in the environment. A.C.S. symposium series 393, Washington, DC, pp 58–69
- Hall J, Reynolds B, Langan S, Hornung M, et al. (2001) Investigating the uncertainties in the simple mass balance equation for acidity critical loads for terrestrial ecosystems in the United Kingdom. *Water Air Soil Pollut Focus* 1:43–56
- Harrison RB, Johnson DW, Todd DE (1989) Sulfate adsorption and desorption reversibility in a variety of forest soils. *J Environ Qual* 18:419–426
- Hicks BB, Baldocchi DD, Meyers TP et al (1987) A preliminary multiple resistance routine for deriving dry deposition velocities from measured quantities. *Water Air Soil Pollut* 36:311–330
- Hicks BB, Hosker RP, Meyers TP et al (1991) Dry deposition inferential measurement techniques. I. Design and tests of a prototype meteorological and chemical system for determining dry deposition. *Atmos Environ* 25:2345–2359
- Houle D, Carignan R (1992) Sulfur distribution and speciation in soils and aboveground biomass of a boreal coniferous forest. *Biogeochemistry* 16:63–82
- Houle D, Carignan R (1995) Role of SO<sub>4</sub> adsorption and desorption in the long-term S budget of a coniferous catchment on the Canadian Shield. *Biogeochemistry* 28:162–182
- Houle D, Paquin R, Camiré C et al (1997) Response of the Lake Clair Watershed (Duchesnay, Québec) to change in precipitation chemistry (1988–1994). *Can J For Res* 27:1813–1821
- Houle D, Carignan R, Ouimet R (2001) Soil organic sulfur dynamics in a coniferous forest. *Biogeochemistry* 53:105–124
- Houle D, Gagnon C, Couture S et al (2004) Recent recovery of lake water quality in southern Quebec following reductions in sulfur emissions. *Water Air Soil Pollut Focus* 4:247–261
- Huntington TG, Hooper RP, Aulenbach BT (1994) Hydrologic processes controlling sulfur mobility: a small watershed approach. *Water Resour Res* 30:283–295
- Huntington TG, Richardson AD, McGuire K et al (2009) Climate and hydrological changes in the northeastern United States: recent trends and implications for forested and aquatic ecosystems. *Can J For Res* 39:199–212
- Inamdar SP, Christopher SF, Mitchell MJ (2004) Export mechanisms for dissolved organic carbon and nitrate during summer storm events in a glaciated forested catchment in New York, USA. *Hydrol Proc* 18:2651–2661
- IPCC (2001) Chapter 4. Atmospheric chemistry and greenhouse gases. In: Houghton JT, Ding Y, Griggs DJ, Noguer M, van der Linden PJ, Dai X, Johnson CA, Maskell K (eds) Climate change 2001: a scientific basis, intergovernmental panel on climate change. Cambridge University Press, Cambridge
- Ito M, Mitchell MJ, Driscoll CT (2002) Spatial patterns of precipitation quantity and chemistry and air temperature in the Adirondack Region of New York. *Atmos Environ* 36:1051–1062
- Jeffries DS, Ouimet R (eds) (2005) Chapter 8: Critical loads— are they being exceeded? In: 2004 Canadian acid deposition science assessment, Environment Canada, Ottawa, ON, pp 341–368. [http://www.msc-smc.gc.ca/saib/acid/assessment2004/pg41\\_e.html#1](http://www.msc-smc.gc.ca/saib/acid/assessment2004/pg41_e.html#1)
- Jeffries DS, Kelso JRM, Morrison IK (1988) Physical, chemical, and biological characteristics of the Turkey Lakes watershed, central Ontario. *Can J Fish Aquat Sci* 45(Suppl 1): 3–12
- Jeffries DS, Clair TA, Couture S et al (2003a) Assessing the recovery of lakes in southeastern Canada from the effects of acidic deposition. *Ambio* 32:176–182
- Jeffries DS, Brydges TG, Dillon PJ et al (2003b) Monitoring the results of Canada/U.S.A. acid rain control programs: some lake responses. *Environ Monit Assess* 88:3–19
- Johnson DW, Mitchell MJ (1998) Responses of forest ecosystems to changing sulfur inputs. In: Maynard D (ed) Sulfur in the environment. Marcel Dekker, Inc, New York, pp 219–262
- Johnson DW, Henderson GS, Huff DD et al (1982) Cycling of organic and inorganic sulphur in a chestnut oak forest. *Oecologia* 54:141–148
- Khana P, Prenzel J, Meiwe KJ et al (1987) Dynamics of sulfate retention by acid forest soils in an acidic environment. *Soil Sci Soc Am J* 51:446–452
- LaZerte BD (1993) The impact of drought and acidification on the chemical exports from a minerotrophic conifer swamp. *Biogeochemistry* 18:153–175
- Likens GE, Bormann FH (1995) *Biogeochemistry of a forested ecosystem*, 2nd edn. Springer-Verlag, New York, pp 159
- Likens GE, Bormann FH, Hedin LO et al (1990) Dry deposition of sulfur: a 23-yr record for the Hubbard Brook Forest Ecosystem. *Tellus* 42B:319–329
- Likens GE, Driscoll CT, Buso DC (1996) Long-term effects of acid rain: response and recovery of a forest ecosystem. *Science* 272:244–246
- Likens GE, Driscoll CT, Buso DC et al (1998) The biogeochemistry of calcium at Hubbard Brook. *Biogeochemistry* 41:89–173

- Likens GE, Butler TJ, Buso DC (2001) Long- and short-term changes in sulfate deposition: effects of the 1990 Clean Air Act Amendments. *Biogeochemistry* 52:1–11
- Likens GE, Driscoll CT, Buso DC et al (2002) The biogeochemistry of sulfur at Hubbard Brook. *Biogeochemistry* 60:235–316
- Likens GE, Buso DC, Butler TJ (2005) Long-term relationships between  $\text{SO}_2$  and  $\text{NO}_x$  emissions and  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  concentration in bulk deposition at the Hubbard Brook Experimental Forest, New Hampshire. *J Environ Monit* 7(10):964–968
- Lovett GM (1984) Rates and mechanisms of cloud water deposition to a sub-alpine balsam fir forest. *Atmos Environ* 18:361–371
- Lovett GM (1994) Atmospheric deposition of nutrients and pollutants in North-America—an ecological perspective. *Ecol Appl* 4:629–650
- Lovett GM, Reiners WA, Olsen RK (1982) Cloud droplet deposition in sub-alpine balsam fir forests: hydrological and chemical inputs. *Science* 218:1303–1304
- Lükewille A, Malessa V, Alewell C (1996) Measured and modelled retention of inorganic sulfur in soils and bedrock (Harz Mountains, Germany). *Water Air Soil Pollut* 85:683–688
- Lynch JA, Bowersox VC, Grimm JW (1996) Trends in precipitation chemistry in the United States, 1983–94: an analysis of the effects in 1995 of phase I of the Clean Air Act Amendments of 1990, Title IV U.S. Geological Survey USGS 96-0346, Washington, DC, June 1996
- Lynch JA, Bowersox VC, Grimm JW (2000) Changes in sulfate deposition in eastern USA following implementation of Phase I of Title IV of the Clean Air Act Amendments of 1990. *Atmos Environ* 34:1665–1680
- Martin CW, Likens GE, Buso DC (2000) Comparison of long-term precipitation chemistry measurements at the Hubbard Brook Experimental Forest, New Hampshire. *Water Air Soil Pollut* 120(3–4):359–379
- Mayer B, Feger KH, Gieseman A et al (1995) Interpretation of sulfur cycling in two catchments in the Black Forest (Germany) using stable sulfur and oxygen isotope data. *Biogeochemistry* 30:51
- Meyers TP, Finkelstein P, Clarke J et al (1998) A multilayer model for inferring dry deposition using standard meteorological measurements. *J Geophys Res Atmos* 103:22645–22661
- Mitchell MJ, Alewell C (2007) Sulfur transformations and fluxes. In: Chesworth W (ed) *The encyclopedia of soil science*. Springer-Verlag, Dordrecht, pp 757–764
- Mitchell MJ, David MB, Maynard DG et al (1986) Sulfur constituents in soils and streams of a watershed in the Rocky Mountains of Alberta. *Can J For Res* 16: 315–320
- Mitchell MJ, Driscoll CT, Fuller RD et al (1989) Effect of whole-tree harvesting on the sulfur dynamics of a forest soil. *Soil Sci Soc Am J* 53:933–940
- Mitchell MJ, David MB, Harrison RB (1992a) Sulfur dynamics of forest ecosystems. In: Howarth RW, Stewart JWB, Ivanov MV (eds) *Sulfur cycling on the continents*. SCOPE vol 48. Wiley, New York
- Mitchell MJ, Foster NW, Shepard JP et al (1992b) Nutrient cycling in Huntington Forest and Turkey Lakes deciduous stands: nitrogen and sulfur. *Can J For Res* 22:457–464
- Mitchell MJ, Mayer B, Bailey SW et al (2001a) Use of stable isotope ratios for evaluating sulfur sources and losses at the Hubbard Brook Experimental Forest. In: *Proceedings of acid rain 2000, Japan*. *Water Air Soil Pollut* 130:75–86
- Mitchell MJ, McHale PJ, Inamdar S et al (2001b) Role of within lake processes and hydrobiogeochemical changes over 16 years in a watershed in the Adirondack Mountains of New York State, U.S.A. *Hydrol Proc* 15:1951–1965
- Mitchell MJ, Piatek KB, Christopher S et al (2006) Solute sources in stream water during consecutive fall storms in a northern hardwood forest watershed: a combined hydrological, chemical and isotopic approach. *Biogeochemistry* 78:217–246
- Mitchell MJ, Bailey SW, Shanley JB et al (2008) Evaluating storm events for three watersheds in the Northeastern United States: a combined hydrological, chemical and isotopic approach. *Hydrol Proc* 22:4023–4034
- Moldan B, Cerny J (eds) (1995) *Biogeochemistry of small catchments: a tool for environmental research*. SCOPE 51. Wiley, Chichester, 419 pp
- Moran MD, Zheng Q, Pavlovic R et al (2008) Predicted acid deposition critical-load exceedances across Canada from a one-year simulation with a regional particulate-matter model. In: *Proceedings of the 15th joint AMS/A&WMA conference on applications of air pollution meteorology*, Jan 21–24, New Orleans, American Meteorological Society, Boston, 20 pp. Available from weblink <http://ams.confex.com/ams/pdfpapers/132916.pdf>
- Morrison IK, Foster NW (2001) Fifteen-year change in forest floor organic and element content and cycling at the Turkey Lakes Watershed. *Ecosystems* 4:545–555
- Morrison IK, Foster NW, Nicolson JA (1992) Influence of acid deposition on element cycling in mature sugar maple forest, Algoma, Canada. *Water Air Soil Pollut* 61:243–252
- Murdoch PS, Shanley JB (2006) Detection of water quality trends at high, median, and low flow in a Catskill Mountain stream, New York, through a new statistical method. *Water Resour Res* 42:W08407. doi:10.1029/2004WR003892
- Nilsson J, Grennfelt P (1988) Critical loads for sulphur and nitrogen: report from a workshop held at Stockolster, Sweden, 19–24 March 1988. Miljörapport 15, Nordic Council of Ministers, 418 pp
- Nodvin SC, Driscoll CT, Likens GE (1986) The effect of pH on sulfate adsorption in a forest soil. *Soil Sci* 142:69–75
- Nordstrom SA (1982) The effect of sulfate on aluminum concentrations in natural waters: stability relations in the system  $\text{Al}_2\text{O}_3\text{--SO}_3\text{--H}_2\text{O}$  at 298 K. *Geochim Cosmochim Acta* 46:681–692
- Norton SA, Fernandez IJ (eds) (1999) *The Bear Brook Watershed in Maine (BBWM)—a paired watershed experiment: the first decade (1987–1997)*. Kluwer Publication, Special Volume (55) of *Environ Monit Assess*
- Novák M, Kirchner JW, Fottova D et al (2005) Isotopic evidence for processes of sulfur retention/release in 13 forested catchments spanning a strong pollution gradient (Czech Republic, Central Europe). *Global Biogeochem Cycles* 19: Art No GB4012
- Novák M, Mitchell MJ, Jacková I et al (2007) Processes affecting oxygen isotope ratios of atmospheric and ecosystem sulfate in two contrasting forest catchments in Central Europe. *ES&T* 41:703–709. doi:10.1021/es0610028

- Ouimet R, Arp PA, Watmough SA et al (2006) Determination and mapping critical loads of acidity and exceedances for upland forest soils in Eastern Canada. *Water Air Soil Pollut* 172:57–66
- Park J, Mitchell MJ, McHale PJ et al (2003) Interactive effects of changing climate and atmospheric deposition on N and S biogeochemistry in a forested watershed of the Adirondack Mountains, New York State. *Global Change Biol* 9:1602–1619
- Piatek KB, Christopher SF, Mitchell MJ (2009) Spatial and temporal dynamics of stream chemistry in a forested watershed. *Hydrol Earth Syst Sci* 13:423–439
- Pleim JE, Xiu A, Finkelstein PL et al (2001) A coupled land-surface and dry deposition model and comparison to field measurements of surface heat, moisture, and ozone fluxes. *Water Air Soil Pollut Focus* 1:243–252
- Porter E, Blett T, Potter DU, Huber C (2005) Protecting resources on federal lands: implications of critical loads for atmospheric deposition of nitrogen and sulfur. *Bio-science* 55:603–612
- Prechtel A, Alewell C, Armbruster M et al (2001) Response of sulphur dynamics in European catchments to decreasing sulphate deposition. *Hydrol Earth Syst Sci* 5:311–325
- Reuss JO, Johnson DW (1986) Acid deposition and acidification of soils and waters. Springer-Verlag, New York, p 119
- Richter DD, Lindberg SE (1988) Wet deposition estimates from long-term bulk and event wet-only samples of incident precipitation and throughfall. *J Environ Qual* 17:619–622
- Rochelle BP, Church MR, David MB (1987) Sulfur retention at intensively studied sites in the U.S. and Canada. *Water Air Soil Pollut* 33:73–83
- Schiff SL, Spoestra J, Semkin RG et al (2005) Drought induced pulses of  $\text{SO}_4^{2-}$  from a Canadian shield wetland: use of  $^{34}\text{S}$  and  $^{18}\text{O}$  in  $\text{SO}_4^{2-}$  to determine sources of sulfur. *Appl Geochem* 20:691–700
- Seip HM, Seip R, Dillon PJ et al (1985) Model of sulphate concentration in a small stream in the Harp Lake catchment, Ontario. *Can J Fish Aquat Sci* 42:927–937
- Shanley JB, Mayer B, Mitchell MJ et al (2005) Tracing sources of streamwater sulfate during snowmelt using S and O isotope ratios of sulfate and  $^{35}\text{S}$  activity. *Biogeochemistry* 76:161–185
- Shanley JB, Mayer B, Mitchell MJ et al (2008) Annual and event variations in  $^{34}\text{S}$  values of stream sulfate in a Vermont forested catchment: implications for S sources and cycling. *Sci Total Environ* 604:262–268
- Shepard JP, Mitchell MJ, Scott TJ et al (1989) Measurements of wet and dry deposition in a northern hardwood forest. *Water Air Soil Pollut* 48:225–238
- Sickles JE, Grimm JW (2003) Wet deposition from clouds and precipitation in three high-elevation regions of the Eastern United States. *Atmos Environ* 37:277–288
- Sickles JE, Shadwick DS (2007) Changes in air quality and atmospheric deposition in the eastern United States: 1990–2004. *J Geophys Res* 112:D17301. doi:10.1029/2006JD007843
- Smyth SC, Jiang W, Roth H et al (2007) A comparative performance evaluation of the AURAMS and CMAQ air quality modelling systems—revised, National Research Council, Ottawa, ON, Canada. ICPET technical report. PET-1577-07S, 114 pp
- Smyth SC, Jiang W, Roth H et al (2008) A comparative performance evaluation of the AURAMS and CMAQ air quality modelling systems. *Atmos Environ* 43:1059–1070
- Staelens J, De Schrijver A, Van Avermaet P et al (2005) A comparison of bulk and wet-only deposition at two sites in Melle (Belgium). *Atmos Environ* 39:7–15
- Stam AC, Mitchell MJ, Krouse HR et al (1992) Dynamics of stable sulfur isotopes of sulfate in precipitation and stream solutions in a northern hardwoods watershed. *Water Resour Res* 28:231–236
- Stoddard JL, Murdoch PS (1991) Catskill mountains. In: Charles DF (ed) *Acidic deposition and aquatic ecosystems*. Springer-Verlag New York Inc, New York, pp 237–271
- Stoddard JL, Jeffries DS, Lukewille A et al (1999) Regional trends in aquatic recovery from acidification in North America and Europe. *Nature* 401:575–577
- Stoddard JL, Kahl JS, Deviney F et al (2003) Response of surface water chemistry to the Clean Air Act Amendments of 1990. EPA/620/R-03/001, U.S. Environmental Protection Agency, Corvallis, OR
- Sullivan TJ, Cosby BJ (2005) Testing, improvement, and confirmation of a watershed model of acid-base chemistry. *Water Air Soil Pollut* 85:2607–2612
- Sullivan TJ, Fernandez IJ, Herlithy AT et al (2006) Acid-base characteristics of soil in the Adirondack Mountains, New York. *Soil Sci Soc Am J* 70:141–152
- Sverdrup H, de Vries W (1994) Calculating critical loads for acidity with the simple mass balance equation. *Water Air Soil Pollut* 72:143–162
- Vet R, Ro C-U (2008) Contribution of Canada–United States transboundary transport to wet deposition of sulphur and nitrogen oxides—a mass balance approach. *Atmos Environ* 42:2518–2529
- Vet R, Lear G, Narayan J et al (2005) Sulfur and nitrogen deposition across eastern North America. In: Abstract from Ecological Society of America meeting, August 9, 2005. Montreal, Canada
- Watmough SA, Aherne J, Alewell C et al (2005) Sulphate, nitrogen and base cation budgets at 21 forested catchments in Canada, the United States and Europe. *Environ Monit Assess* 109:1–36
- Weathers KC, Lovett GM (1998) Acid deposition research and ecosystem science: synergistic successes. In: Pace ML, Groffman PM (eds) *Successes, limitations, and frontiers in ecosystem science VII Cary conference*. Springer-Verlag, New York, pp 195–219
- Weathers KC, Lovett GM, Likens GE (1995) Cloud deposition to a spruce forest edge. *Atmos Environ* 29:665–672
- Weathers KC, Lovett GM, Likens GE et al (2000) The effect of landscape features on deposition to Hunter Mountain, Catskill Mountains, New York. *Ecol Appl* 10:528–540
- Weathers KC, Butler TJ, Likens GE (2006a) Acid rain. In: Rom WN (ed) *Environmental and occupational medicine*. Lippincott, Williams, and Wilson, Philadelphia, pp 1507–1520
- Weathers KC, Simkin SM, Lovett SM et al (2006b) Empirical modeling of atmospheric deposition in mountainous landscapes. *Ecol Appl* 16:1590–1607
- Yanni S, Keys K, Clair TA et al (2000) Fog and acidification impacts on ion budgets of basins in Nova Scotia, Canada. *J Am Water Resour Assoc* 36:619–631

Zhang Y, Mitchell MJ, Christ M et al (1998) Stable sulfur isotopes at Hubbard Brook Experimental Forest, New Hampshire. *Biogeochemistry* 41:259–275

Zhang L, Brook JR, Vet R et al (2001) Evaluation and Improvement of a dry deposition model using SO<sub>2</sub> and O<sub>3</sub>

measurements over a mixed forest. *Water Air Soil Pollut Focus* 1:67–78

Zhang L, Vet R, O'Brien JM et al (2009) Dry deposition of individual nitrogen species at eight Canadian rural sites. *J Geophys Res* 114:DO2301. doi:[10.1029/2008JD010640](https://doi.org/10.1029/2008JD010640)